

Hydrogen atom

electron has orbital angular momentum $\vec{L}$
and spin angular momentum $\vec{S}$

Total angular momentum $\vec{J} = \vec{L} + \vec{S}$

Product space state

$$|l, m_l; s, m_s\rangle \text{ or } |m_l; m_s\rangle$$

Direct sum state

$$|j, m_j\rangle$$

l	$\text{Spin } l \otimes (\text{spin } \frac{1}{2})$	<u>multiplicities</u>	$\text{Spin } (l + \frac{1}{2}) \oplus \text{Spin } (l - \frac{1}{2})$
0 (s)	$\underline{1} \otimes \underline{2}$	2	$\underline{2} \quad S_{\frac{1}{2}}$
1 (p)	$\underline{3} \otimes \underline{2}$	6	$\underline{4} \oplus \underline{2} \quad P_{\frac{3}{2}}, P_{\frac{1}{2}}$
2 (d)	$\underline{5} \otimes \underline{2}$	10	$\underline{6} \oplus \underline{4} \quad d_{\frac{5}{2}}, d_{\frac{3}{2}}$
3 (f)	$\underline{7} \otimes \underline{2}$	14	$\underline{8} \oplus \underline{6} \quad f_{\frac{7}{2}}, f_{\frac{5}{2}}$
$\underbrace{\hspace{10em}}$ eigenstates of L_z, S_z			$\underbrace{\hspace{10em}}$ (subscript labels j) eigenstates of J^2, J_z

eg.

$$n=2$$

$l=0$

$$\frac{2s}{102}$$

$l=1$

$$\frac{2p}{302}$$

$\Rightarrow$

$$\frac{2s_{1/2}}{\underline{2}}$$

$$\frac{2p_{1/2}}{\underline{2}}$$

$$\frac{2p_{3/2}}{\underline{4}}$$

The non-relativistic Schrodinger eqn for hydrogen
has energy levels $E_n = -\left(\frac{mc^2\alpha^2}{2}\right) \frac{1}{n^2}$,
as we'll see.

$$\alpha = \frac{Ke^2}{\hbar c}$$

↑
"fine structure constant"

There are relativistic and spin-orbit corrections
to the Hamiltonian (see early problem set) → (Dirac eqn)

$$\delta E_{\text{rel}} = -\frac{p^4}{8m^3c^2}$$

$$\delta E_{\text{S.O.}} = \frac{Ke^2}{2m^2c^2r^3} \vec{L} \cdot \vec{S} = \frac{mc^2\alpha^4}{2} \left(\frac{a_0}{r}\right)^3 \frac{\vec{L} \cdot \vec{S}}{\hbar^2}$$

which give corrections of $O(\alpha^2)$ to the energy levels
(called fine structure corrections).

We can't yet calculate the effect of p^4 and $\frac{1}{r}$

Before, we estimated $\vec{L} \cdot \vec{S} \approx \frac{1}{2}\hbar^2$

now do it more carefully

observe that $\vec{J} = \vec{L} + \vec{S}$

$$J^2 = L^2 + S^2 + 2\vec{L} \cdot \vec{S}$$

$$\vec{L} \cdot \vec{S} = \frac{1}{2}(J^2 - L^2 - S^2)$$

direct sum states are $\therefore$ eigenstates of $\vec{L} \cdot \vec{S}$

$$[L^2, L_z] + [S^2, S_z] + 2[L_z, S_z]$$

Let $\vec{L} \cdot \vec{S}$ act on states of total angular momentum $|j, m_j\rangle$

$$\vec{L} \cdot \vec{S} |j, m_j\rangle = \frac{\hbar^2}{2} [j(j+1) - l(l+1) - \frac{1}{2}(\frac{1}{2}+1)] |j, m_j\rangle$$

$$\text{For } j = l + \frac{1}{2} \Rightarrow \frac{\hbar^2}{2} [(l + \frac{1}{2})(l + \frac{3}{2}) - l(l+1) - \frac{3}{4}] = l \frac{\hbar^2}{2}$$

$$\text{For } j = l - \frac{1}{2} \Rightarrow \frac{\hbar^2}{2} [(l - \frac{1}{2})(l + \frac{1}{2}) - l(l+1) - \frac{3}{4}] = (-l-1) \frac{\hbar^2}{2}$$

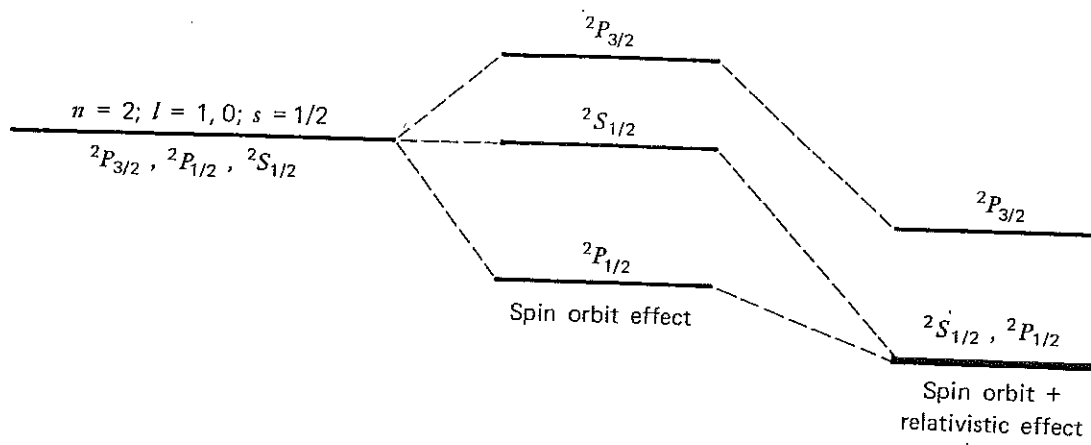
$$\delta E_{so.} = \frac{mc^2 \alpha^4}{4} \left(\frac{a_0}{r}\right)^3 \begin{cases} l & \text{for } j = l + \frac{1}{2} \\ -l-1 & \text{for } j = l - \frac{1}{2} \end{cases}$$

Spin-orbit effect

$s_{\frac{1}{2}}$ state is unchanged

$p_{\frac{1}{2}}$ state goes down

$p_{\frac{3}{2}}$ state goes up



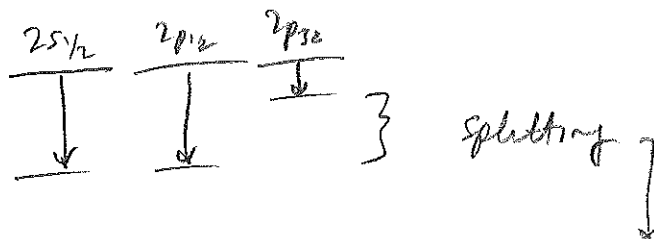
The full fine structure correction

$$\delta E = \delta E_{s.o.} + \delta E_{rel}$$

$$\left[\begin{array}{l} \text{A long} \\ \text{calculation} \\ \text{gives} \end{array} \right] = \frac{mc^2 \alpha^4}{2} \left[\frac{3}{4n^4} - \frac{1}{n^3(j + \frac{1}{2})} \right]$$

$$2s_{1/2}, 2p_{1/2}: \quad \frac{3}{4(2)^4} - \frac{1}{(2)^3(1)} = \frac{3}{64} - \frac{8}{64} = -\frac{5}{64} \quad \text{goes down}$$

$$2p_{3/2}: \quad \frac{3}{4(2)^4} - \frac{1}{(2)^3(2)} = \frac{3}{64} - \frac{4}{64} = -\frac{1}{64} \quad \text{goes down less}$$



split

$$\delta E_{3/2} - \delta E_{1/2} = \frac{mc^2 \alpha^4}{2} \left[-\frac{1}{64} - \left(-\frac{5}{64} \right) \right] = \frac{mc^2 \alpha^2}{2} \left(\frac{\alpha^2}{16} \right)$$

$$= (13.6 \text{ eV}) (3 \times 10^{-6})$$

$$= 4.5 \times 10^{-5} \text{ eV}$$

$$[\text{Lamb shift } \delta E_{p_{1/2}} - \delta E_{s_{1/2}} = 4.3 \times 10^{-6} \text{ eV}]$$

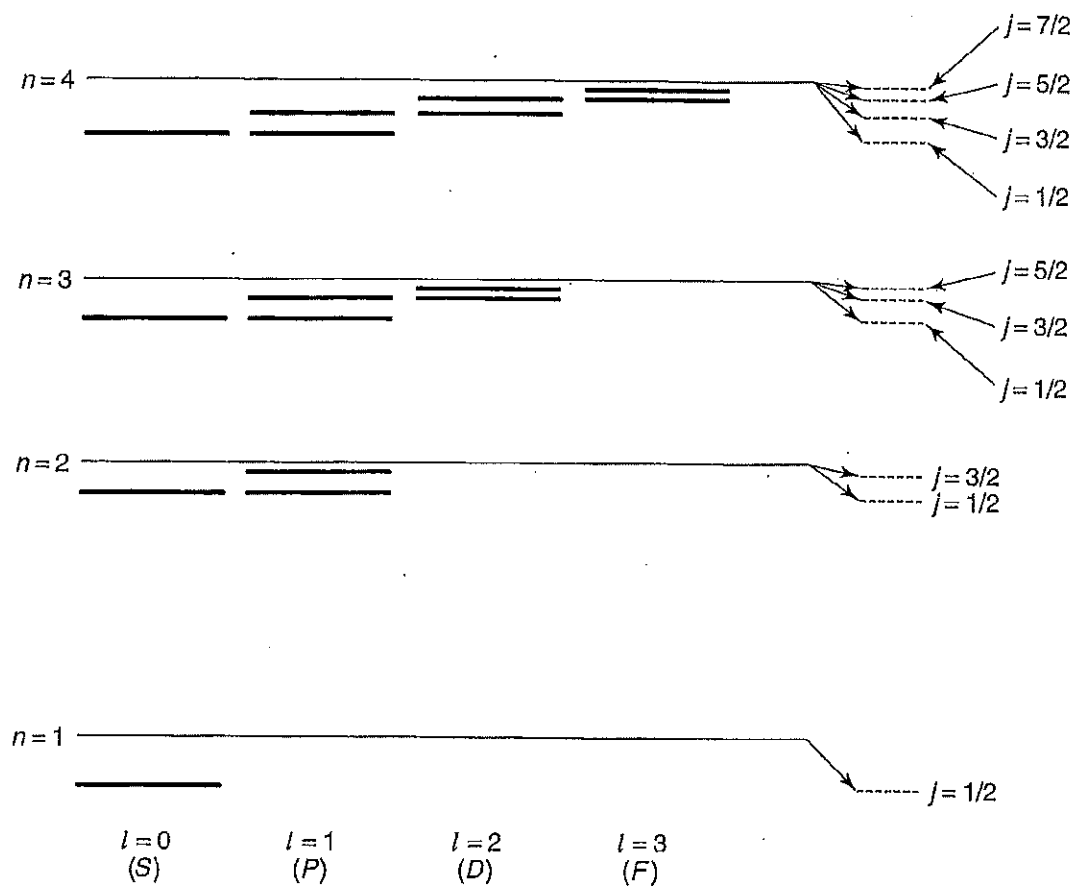


FIGURE 6.9: Energy levels of hydrogen, including fine structure (not to scale).

OLD NOTES

Let's estimate these.

Recall in Bohr model, typical value of $\frac{v}{c}$ is $\alpha \sim \frac{1}{137}$

so typical $p = mv$ is $m c \alpha$

and typical r is Bohr radius $a_0 = \frac{\hbar}{m c \alpha}$ (from $m v a_0 = \hbar$)

and typical $L + S$ is $\hbar$ so

$$\begin{aligned} \delta E &= \frac{K e^2 \hbar^2}{2 m^2 c^2 a_0^3} \left(\frac{a_0}{r}\right)^3 \frac{\vec{L} \cdot \vec{S}}{\hbar^2} \quad \left(\frac{(m c \alpha)^4}{8 m^3 c^2} \left(\frac{\hbar}{m c \alpha}\right)^4 \right) \\ &= \frac{m c^2 \alpha^4}{2} \left(\frac{a_0}{r}\right)^3 \left(\frac{\vec{L} \cdot \vec{S}}{\hbar^2}\right) - \frac{m c^2 \alpha^4}{8} \left(\frac{\hbar}{m c \alpha}\right)^4 \end{aligned}$$

← done in HW

∴ both corrections are of order $m c^2 \alpha^4$

+ suppressed by factor $\alpha^2 \sim 10^{-4}$ wrt. leading term.

These are called fine-structure corrections

+ α = fine-structure constant

[see Griffiths]

Using Schrodinger eqn wavefunctions one finds

→ derived later in the semester

$$\left(\frac{a_0}{r}\right)^3 = \frac{1}{n^3 l(l+\frac{1}{2})(l+1)}$$

$$\left(\frac{\hbar}{m c \alpha}\right)^4 = \frac{4}{n^3(l+\frac{1}{2})} - \frac{3}{n^4}$$

Step 1.

Including both corrections we have

Ans

$$\delta E = \frac{mc^2 \alpha^4}{4n^3 l(l+\frac{1}{2})(l+1)} \left\{ \begin{matrix} l \\ -l-1 \end{matrix} \right\} = \frac{mc^2 \alpha^4}{8} \left[\frac{4}{n^3(l+\frac{1}{2})} - \frac{3}{n^4} \right]$$

all levels go down

[see diagram from]
Bascom

Observe that now $S_{\frac{1}{2}}$ and $P_{\frac{1}{2}}$ have same energy
ie energy depends on j but not l . [for hydrogen]

This is general. w/ some algebra

$$\delta E = \delta E_{r.o.} + \delta E_{rel.}$$

$$\delta E = \frac{mc^2 \alpha^4}{2} \left[\frac{3}{4n^4} - \frac{1}{n^3(j+\frac{1}{2})} \right]$$

A long calculation gives



$$\frac{3}{4(2)^4} - \frac{1}{(2)^3(1)} = \frac{3}{64} - \frac{8}{64} = -\frac{5}{64}$$

$$\frac{3}{4(2)^4} - \frac{1}{(2)^3(2)} = \frac{3}{64} - \frac{4}{64} = -\frac{1}{64}$$

~~talk about
nearby
of
hydrogen~~

Estimate splitting between $np_{3/2}$ and $np_{1/2}$

$$\frac{mc^2 \alpha^4}{2} \left[-\frac{1}{n^3} + \frac{1}{2n^3} \right] = \frac{mc^2 \alpha^2}{2m^2} \left[\frac{\alpha^2}{2n} \right]$$

$$n=2 \Rightarrow (13.6 \text{ eV}) \left(\frac{\alpha^2}{16} \right) = 4.5 \times 10^{-5} \text{ eV}$$

fractional shift

[Lamb shift $2S_{1/2}, 2P_{1/2} \Rightarrow 4.3 \times 10^{-6} \text{ eV}$]