

more history of nuclear physics

(07)-fi-11

Nuclear transmutation

As discussed earlier
Rutherford & collaborators used α particles emitted by Po & Ra

γ typical $T = 5 \text{ MeV}$ to probe atomic structure.

For heavy targets (Z_2 large), Coulomb barrier prevents

α from getting too close to nucleus,

which therefore behaves as a point charge

$$\gamma \frac{d\sigma}{d\Omega} \sim \frac{1}{\sin^4 \frac{\theta}{2}}$$

Scattering from lighter nuclei begins to depart from this
suggesting a finite size for nucleus.

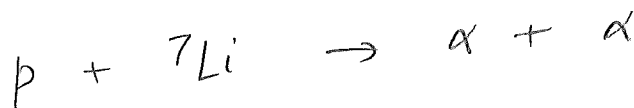
If α gets close enough ($\sim 1 \text{ fm}$), strong force
can cause some of its particles to be absorbed

1919 Nuclear transmutation (Rutherford)

← observed
scintillations
due to p



1932 Cockcroft/Walton [designed a] linear accelerator to
accelerate protons to $\sim 0.5 \text{ MeV}$, and
[with this, they] "split the atom"



20

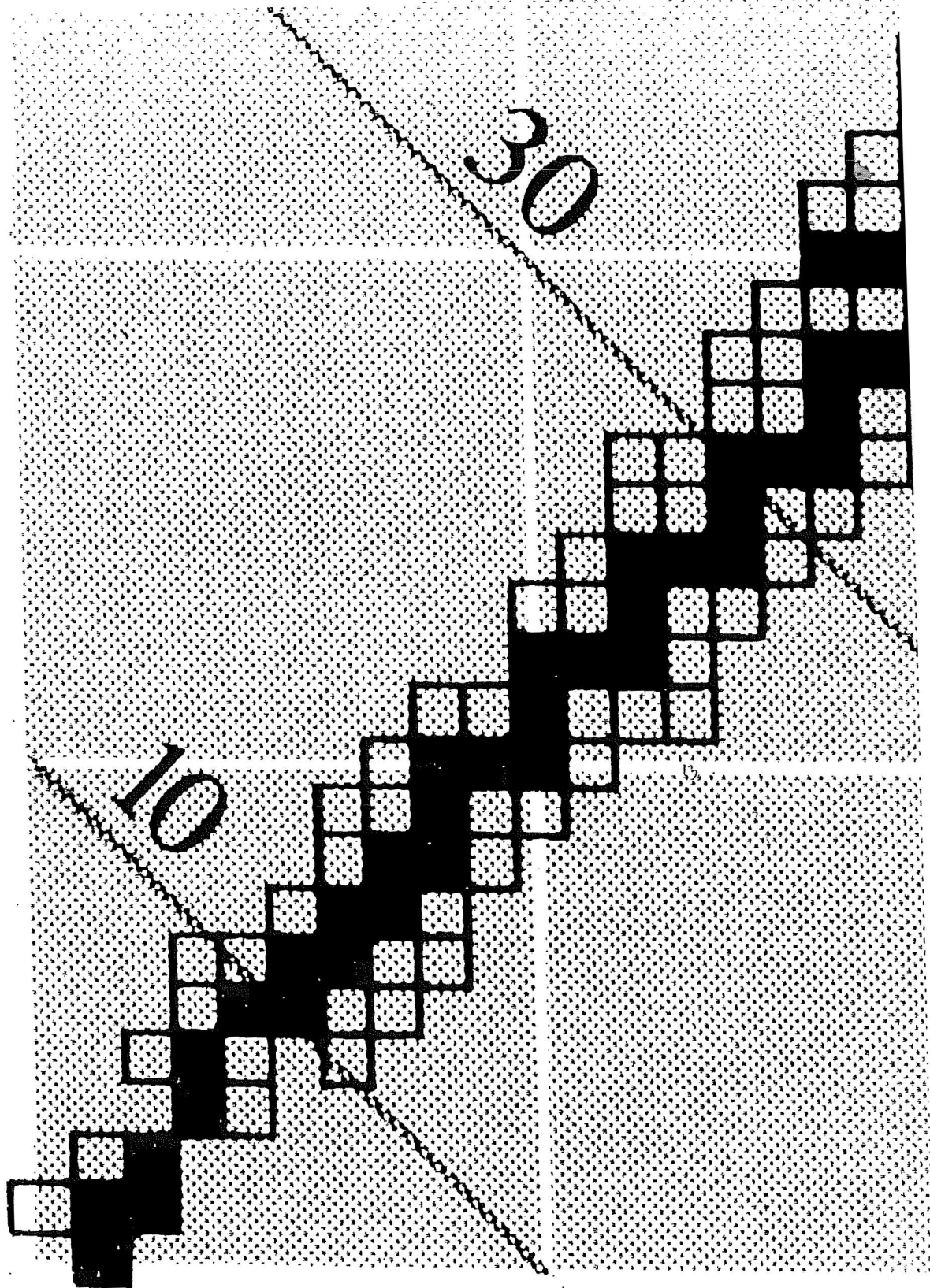
30

10

10

0

10



1932 Chadwick discovers the neutron

neutral particles
knocks protons
out of paraffin



[also Bothe & Becker 1930]

1934 Artificially radioactive isotopes (Joliot & I. Curie)



↓ 3 minutes



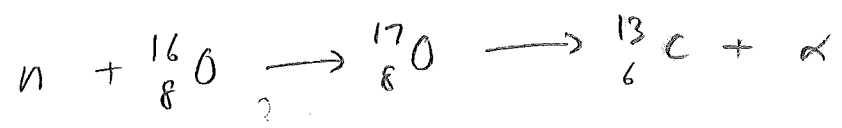
because on proton rich side

[first β^+ decay]

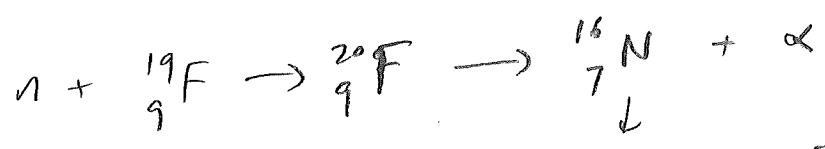
Nobel Chemistry 1935

Neutrons as probes [no Coulomb barrier; but can't accelerate them either]

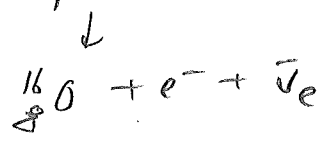
[today
directing
transmutation
tools]



(Meitner & Philipp)

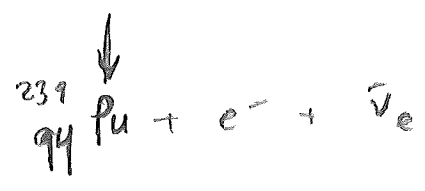
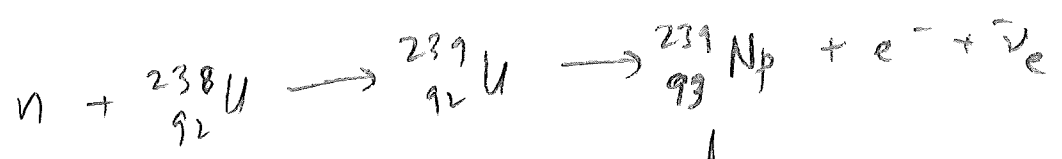


(Fermi & Rasetti)

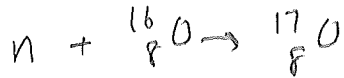


Many artificially radioactive isotopes produced [nothing below fluorine]

even beyond uranium (transuranic)



[international character of physics]



$$Q = 8.071 + (-4.737) - (-0.804)$$

$$= 4.143 \text{ MeV}$$



$$Q = 8.071 + (-4.737) - (3.127) - (2.424)$$

$$= -2.215 \text{ MeV}$$

process require $K_n \geq 2.2 \text{ MeV}$



$$Q = 8.071 + (-1.487) - (5.682) - (2.424)$$

$$= -1.522 \text{ MeV}$$

process require $K_n \geq 1.5 \text{ MeV}$

Note international (or at least European)
nature of this research

Also, one of the physicists, Lise Meitner
was one of the few women doing physics research
at that time.

She was Austrian by birth, educated at U Vienna,
working in Berlin at Kaiser Wilhelm Institute
w/ Otto Hahn (a chemist)

Beside laboring under the disadvantages
suffered by women scientists in that era,
she was also Jewish, which was problematic
in Germany in the 1930's — most German Jewish
scientists had been fired in 1933 after Hitler came
to power.

She was protected by her Austrian citizenship
until 1938, when Germany annexed Austria

In immediate danger, friends helped her to escape
to Netherlands & then to Sweden.

We'll encounter her again later

Neutron absorption by nuclei



$$R = A^{1/3} r_0, \quad r_0 = 1.2 \text{ fm}$$

Geometrical cross-section

$$\sigma = \pi R^2 = \pi A^{2/3} r_0^2$$

E.g., for $A \sim 100$ [Ruthenium],

$$\sigma \sim 100 \text{ fm}^2 = 10^{-24} \text{ cm}^2$$

Absorption cross-section for medium energy ($E \approx 10 \text{ MeV}$) neutrons by nuclei $\approx$ geometric cross section

[See Evans, fig 2-3]

[suggests all neutrons that "hit" the nucleus are absorbed]

[like hitting the broadside of a barn]

$$\text{unit of cross-section} = 1 \text{ barn} = 100 \text{ fm}^2 = (10 \text{ fm})^2$$

[For higher energy neutrons, $\sigma \neq$
so not all neutrons are absorbed
"partial transparency"]

$$[\text{Hw: } ^{238}\text{U} \Rightarrow \sigma = 1.7 \text{ barns}]$$

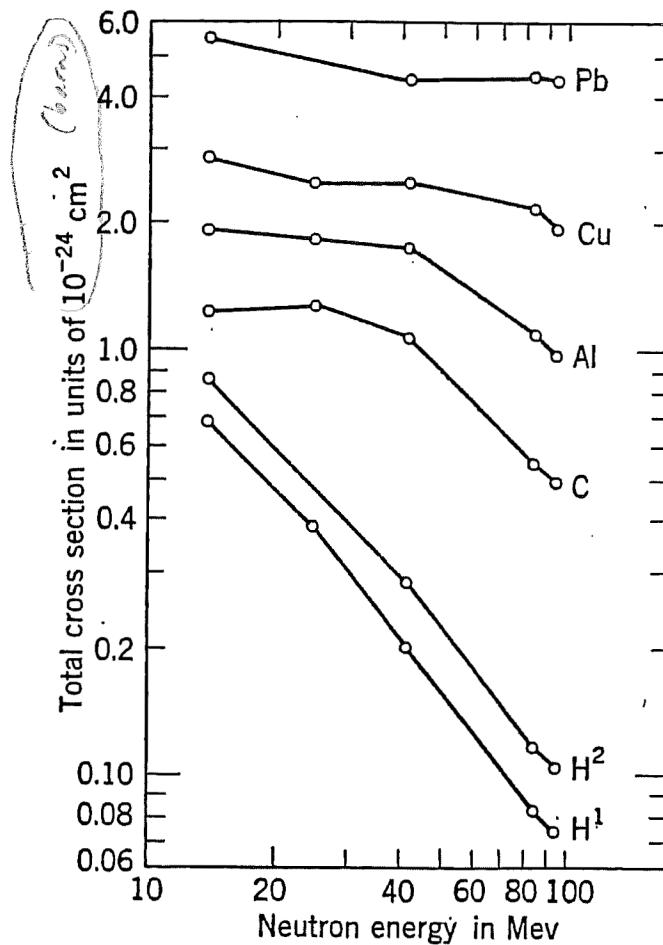
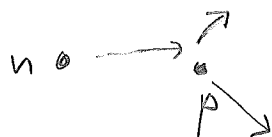


Fig. 2.3 Total neutron cross sections σ_t of five elements for neutrons of 14, 25, 42, 84, and 95 Mev. The decrease in σ_t at the higher neutron energies suggests that all neutrons which strike the target area $\pi(R + \lambda)^2$ are not captured and that nuclear matter is slightly transparent to swift neutrons. [From Hildebrand and Leith (H51).]

Neutrons, being neutral, cannot be accelerated or slowed
 but they can be slowed down by collisions, or by
 absorption + re-emission



Fermi found that absorption $\sigma \uparrow$ for low energy n

[Evan's fig 2.1]

Slow neutrons most likely to be absorbed!

P. minimum ~ 2000 barns for slow neutrons

[empirically $\sigma \sim \frac{1}{v}$]

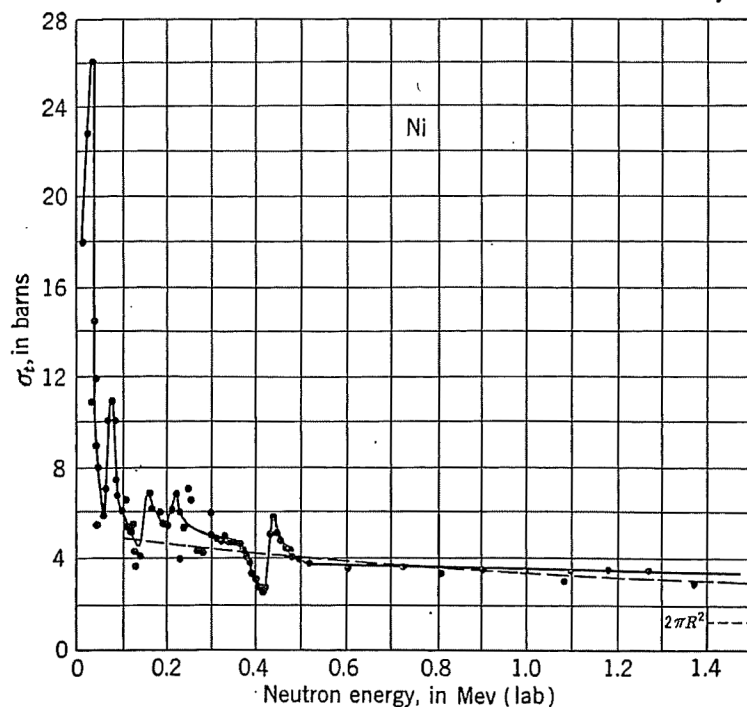


Fig. 2.1 The total cross section (absorption plus scattering) of nickel for neutrons from 0.01 to 1.5 Mev. Approximately monoenergetic neutrons (± 20 kev below 0.5 Mev; ± 150 kev above 0.5 Mev) were produced by bombarding a thin Li target with protons accelerated by an electrostatic generator. The dotted line is from the continuum theory of Feshbach and Weisskopf (F49), assuming a nuclear radius of 4.6×10^{-13} cm. [The experimental data are from Barschall, Bockelman, and Seagondollar (B15).]

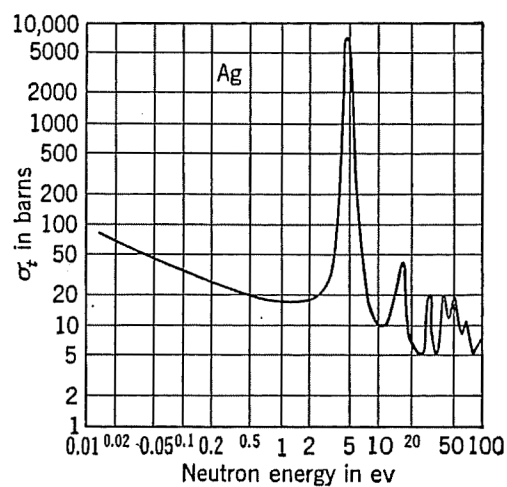
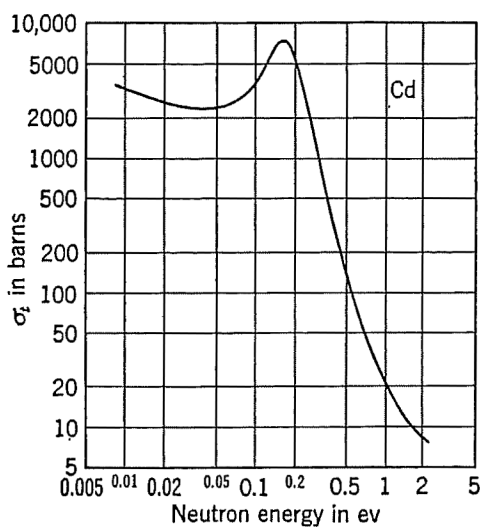
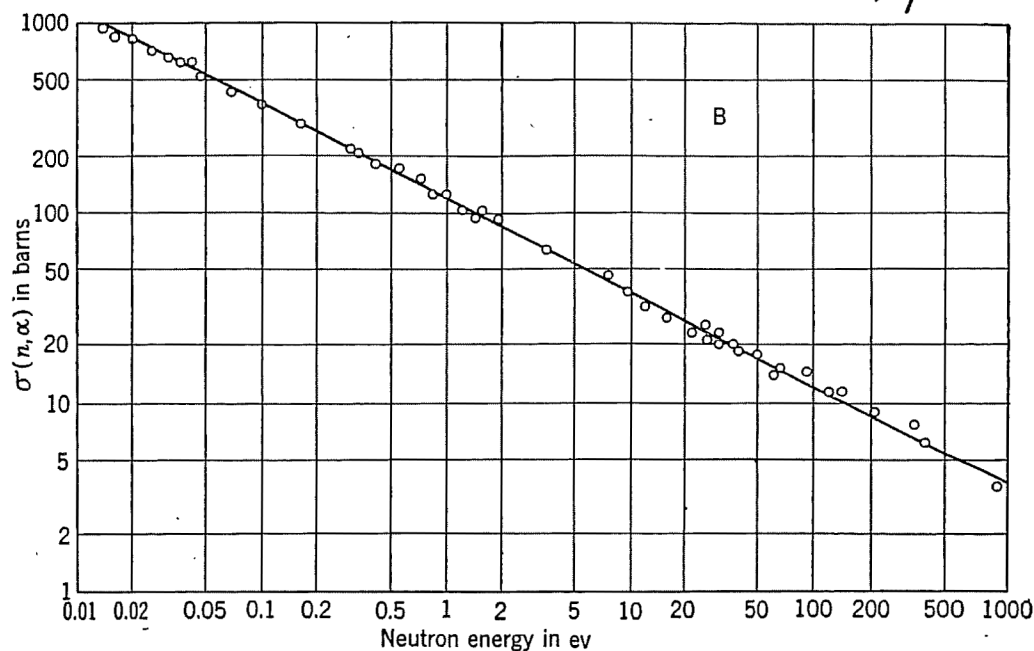


Fig. Introduction.1 Total neutron cross sections for B, Cd, and Ag, illustrating typical combinations of $1/v$ dependence (pure, in the case of boron) and slow-neutron resonances. The resonance in Cd at $E_0 = 0.176$ eV has a total width $\Gamma = 0.115$ eV and a maximum value of $\sigma_0 = 7,200$ barns. The curves are for the normal isotopic mixtures. In Cd, the 0.176-eV resonance is due entirely to Cd^{113} (relative abundance, 12.3 per cent); hence $\sigma_0 \simeq 57,000$ barns for Cd^{113} alone. The complex γ -ray spectrum leading to the ground level of Cd^{114} has been measured by Bartholomew and Kinsey (B16). [From Goldsmith, Ibser, and Feld (G29) and Hughes (H68).]

An old note to self (copied)

How does $\sigma_{fission}$ depend on E of incident n ?

Naively $E \sim v^2 \sim \frac{1}{\lambda^2} \sim \frac{1}{\sigma}$

so $\sigma \sim \frac{1}{E}$

but $\sigma(\text{thermal}) \sim 640 \text{ E-24 cm}^2$

$\sigma(\text{mev}) \sim 1.5 \text{ E-24 cm}^2$

only factor of 500

[skipped in 2019]

P. 11

Why? wave properties of neutrons

de Broglie wavelength $\lambda = \frac{h}{p}$

Define reduced wavelength $\bar{\lambda} = \frac{\lambda}{2\pi} = \frac{h}{p}$

Consider non-relativistic neutron ($T \ll m_n c^2 \sim 1000 \text{ MeV}$)

$$T = \frac{p^2}{2m} = \frac{1}{2m} \left(\frac{h}{\bar{\lambda}} \right)^2 = \frac{1}{2mc^2} \left(\frac{hc}{\bar{\lambda}} \right)^2 = \frac{1}{2(1000 \text{ MeV})} \left(\frac{200 \text{ MeV} \cdot \text{fm}}{\bar{\lambda}} \right)^2$$
$$= (20 \text{ MeV}) \left(\frac{1 \text{ fm}}{\bar{\lambda}} \right)^2$$

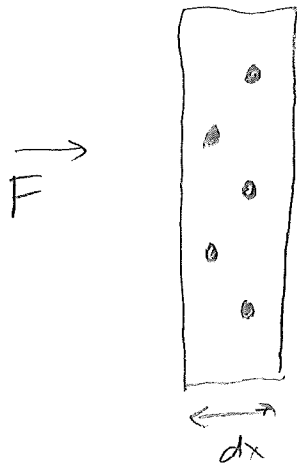
$$\frac{\bar{\lambda}}{1 \text{ fm}} = \sqrt{\frac{20 \text{ MeV}}{T}}$$

For $T \gtrsim 20 \text{ MeV}$, $\bar{\lambda} < 1 \text{ fm}$, neutron acts as a particle
($\bar{\lambda} \ll R$)

Small T , $\bar{\lambda} \gg R$, act like a wave, $\sigma \propto \phi$

Neutrons of $T \sim \frac{3}{2} kT \sim 0.05 \text{ eV}$ are called "thermal neutrons"

Let incident flux of neutrons be F



σ = cross-section for absorption

n = Number density of nuclei

targets $n(A dx)$

"area" of targets $= n\sigma A dx$

Probability of absorption by nuclei in film of thickness dx

$$f = n\sigma(dx)$$

$\Rightarrow$ Flux correspondingly diminishes

$$dF = -(n\sigma dx) F$$

$$\frac{dF}{F} = -n\sigma dx$$

$$\ln F = -n\sigma x + \ln F_0$$

$$F = F_0 e^{-n\sigma x}$$

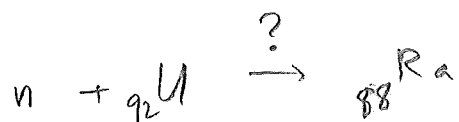
[Hw] show

$$\left[l = \frac{1}{n\sigma} = \text{mean free path of neutrons} \right]$$

$$\left[\begin{aligned} n &= \frac{\rho}{A} N_A & \rho &= \text{mass density} \\ U \rightarrow \rho &= 19 \frac{\text{g}}{\text{cm}^3} & \Rightarrow n &= 4.8 \times 10^{28} \text{ m}^{-3} \\ & & \sigma &= 1.7 \times 10^{-28} \text{ m}^2 \\ & & l &\approx 0.12 \text{ m} \end{aligned} \right]$$

[Hw]

1938 Hahn & Strassmann (chemists)
thought they observed



but no physical process could explain this

- why did they think it was radium?
- Because it had similar chemical properties to Ba and a radioactive product was precipitating out of soln w/ Ba
- But they then tried to separate Ra from Ba and failed
- What did they do? Consulted a physicist.

Former colleague Lise Meitner

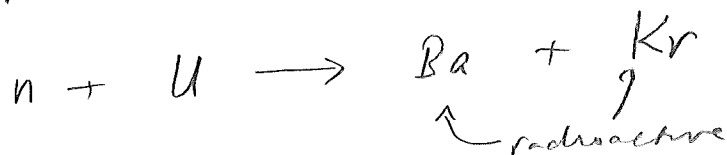
an Austrian Jew.

When Hitler annexed Austria, she lost her citizenship

In 1938, she escaped to Sweden.

She discussed it w/ nephew Otto Frisch and decided it was Ba.

1939 Meitner & Frisch propose



"nuclear fission"

Why didn't anyone think of this before?

$$\frac{B}{A} \text{ for U} \sim 7.5 \text{ MeV}$$

$$\frac{B}{A} \text{ for Ba, Kr} \sim 8.2 \text{ MeV}$$

about 0.7 MeV released per nucleon

$\Rightarrow \sim 170 \text{ MeV per fission.}$

$\Rightarrow$ kinetic energy of fission products

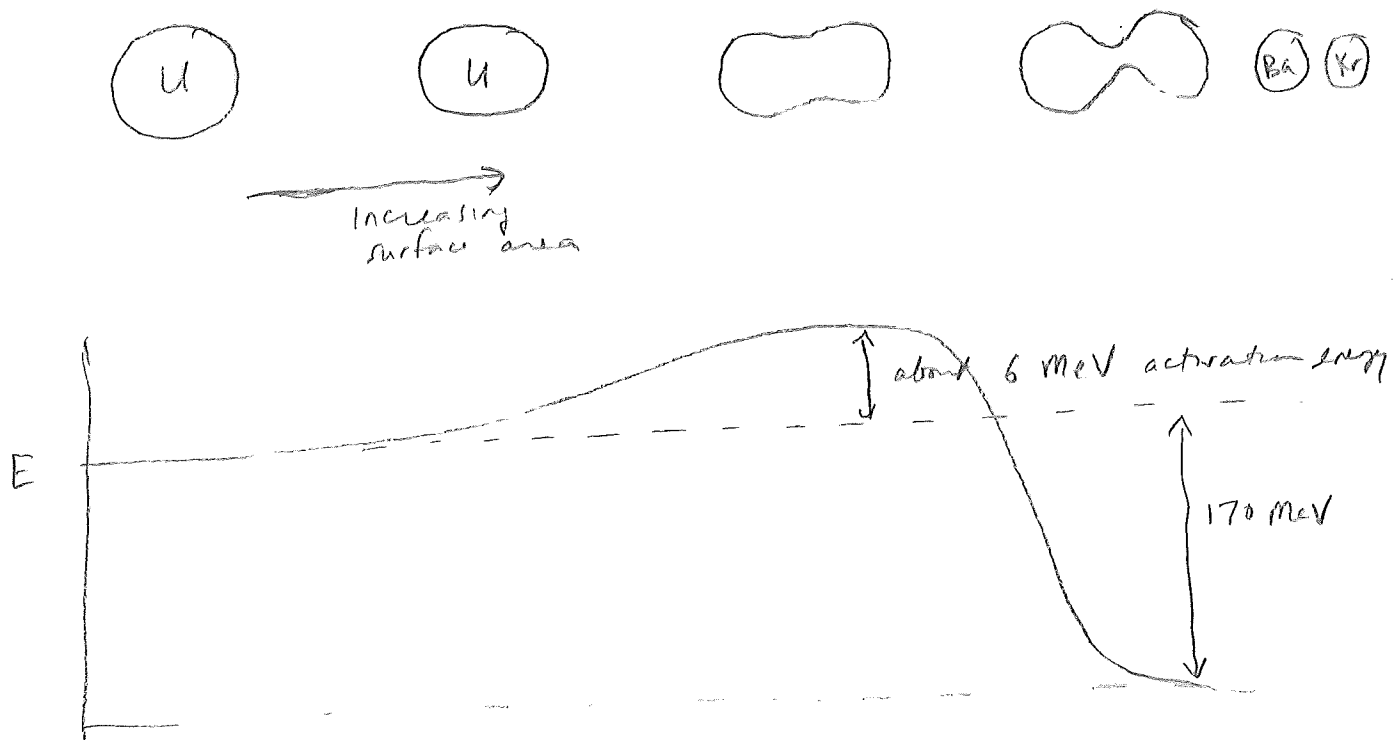
Frisch in his lab verified energy of fragments.

fi-9

If U less stable than Ba, Kr why doesn't it just decay?

Think α -decay. Coulomb barrier holds them in.

Alternative picture. Liquid drop model (surface tension)



*Tunnelling through barrier can occur but is rare
"spontaneous fission"

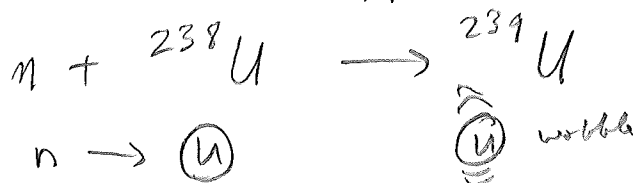
$$T_{\frac{1}{2}} \sim 3.5 \times 10^{17} \text{ years}$$

[HKK, p. 1169]

(α -decay is more likely)

Incident neutron supplies energy

fi-10



$$Q = \Delta({}^{238}\text{U}) + \Delta(n) - \Delta({}^{239}\text{U}) = 4.8 \text{ MeV}$$

$$\begin{bmatrix} 47.306 \\ 8.071 \\ -51.571 \\ \hline 4.806 \end{bmatrix}$$

energy released causes drop to wobble

but activation energy of ${}^{239}\text{U}$ is 6.6 MeV

[? 5.7 MeV in HRL, 1156]

neutrons must have $T > 1.8 \text{ MeV}$ to initiate fission

"fast neutron fission of ${}^{238}\text{U}$ ", $\frac{v}{c} \sim \sqrt{\frac{2T}{mc^2}} \sim 0.06$

But Hahn & Strassmann were using slow (thermal) neutrons ($T < 1 \text{ eV}$)

1939 Bohr realized that they were observing "slow neutron fission of ${}^{235}\text{U}$ "

${}^{238}\text{U}$ 99.3%
 ${}^{235}\text{U}$ 0.7%



$$\begin{bmatrix} 40.916 \\ 8.071 \\ -42.442 \\ \hline 6.545 \end{bmatrix}$$

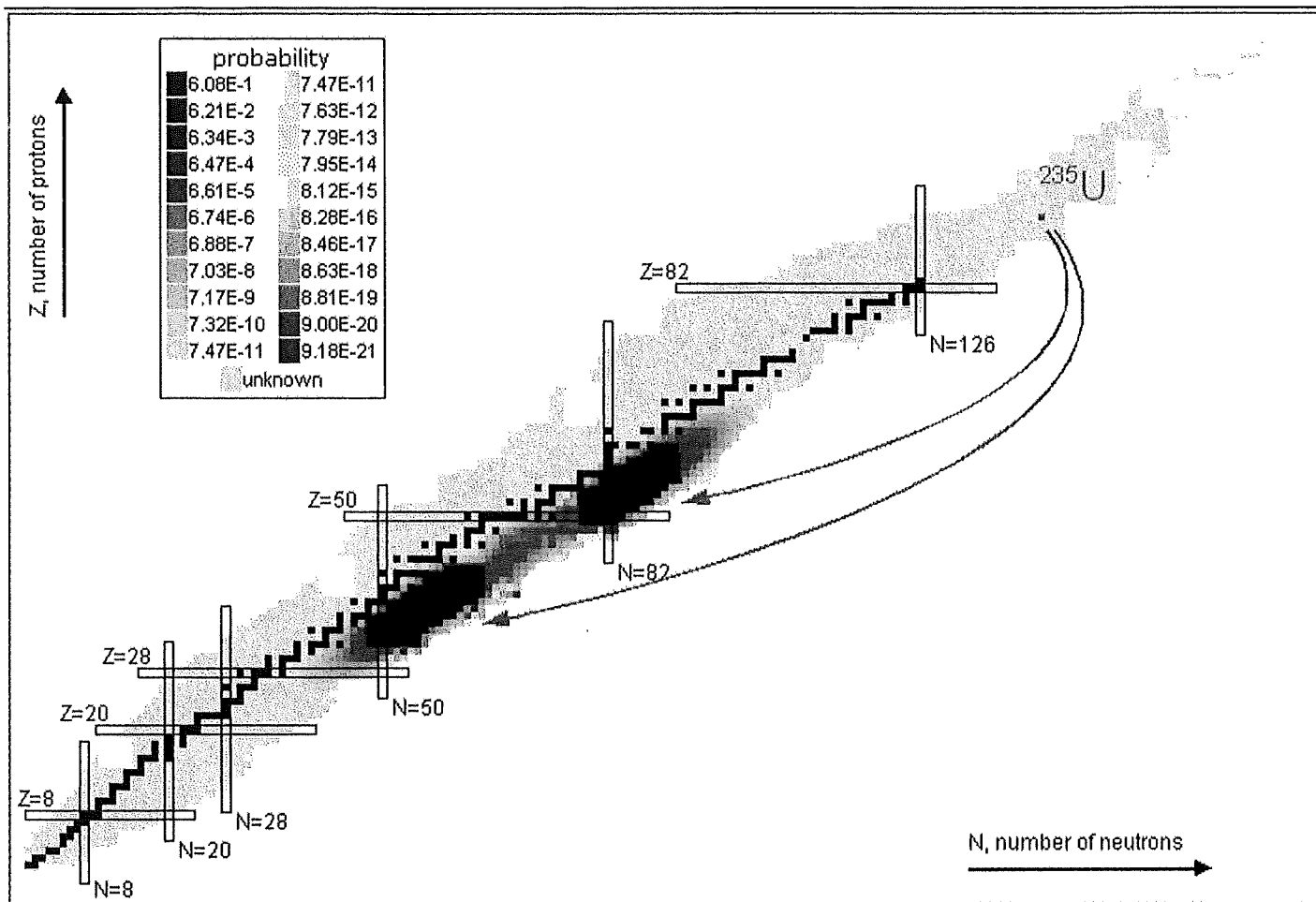
$$Q = 6.5 \text{ MeV}$$

$\Rightarrow$ why difference? even $\rightarrow$ odd, odd $\rightarrow$ even

and activation energy of ${}^{236}\text{U}$ is 6.2 MeV

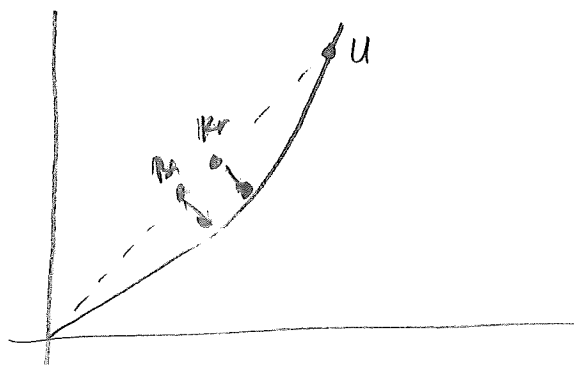
no additional kinetic energy required!

[? 5.2 in HRL]



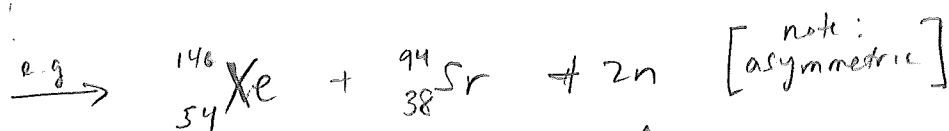
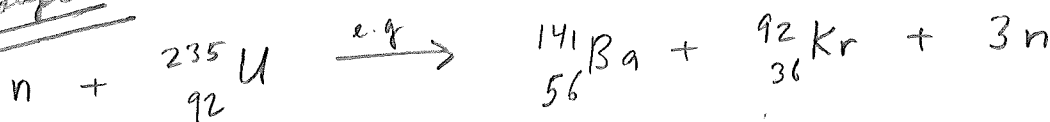
Lab Deming Lounge ~~Hyper~~
blogspot.com

[Why are Ba + Kr radioactive? Because they are neutron rich]
 ↓
 [think about it more]



Fission often releases 2 or 3 neutrons in addition to fission fragments

For example



Released neutrons can initiate further fission

Leo Szilard $\Rightarrow$ chain reaction

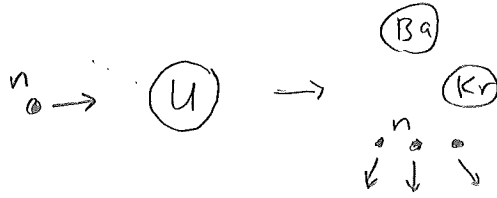
controlled $\Rightarrow$ nuclear reactor
 uncontrolled $\Rightarrow$ weapons

about 11.6 MeV released per fission
 vs 1-10 eV released in chemical reactions $\Rightarrow$ factor of 10^7 to 10^8

Leo Szilard convinced Einstein to write to Roosevelt...

First controlled nuclear reaction (Dec 2, 1942)

U of Chicago squash courts (Fermi)



Secondary neutrons,
avg $H \sim 2.47$ /fission
avg $T \sim 2$ MeV
(some are delayed)

escape through surface
w/o causing fission

↓ solution
Increase volume +
surface ratio
(critical mass)
↓ depends on
material

fast n fission
of ${}^{238}\text{U}$

[possible since $\bar{E} > 1.8$ MeV]

occurs $< 20\%$ of time
[because of low cross-section
for fast n absorption]
→ not sustainable
[need at least 400%]

slowed by
inelastic collisions
w/ ${}^{238}\text{U}$
($> 80\%$ of time)

captured
re-emitted
if $E < 1$ MeV

slow neutrons
absorbed by
 ${}^{238}\text{U} \rightarrow {}^{239}\text{U}$
% subsequent
fission
(Between
1-100 eV)
capture resonance

slowed by
a moderator
(water,
heavy water,
ultra-pure
graphite)
below 1 eV

slow n fission
of ${}^{235}\text{U}$
thermal
neutrons
(0.04 eV)

For fast and rxn, need

- 1) critical mass,
- 2) pure moderator

fast k - multiplication factor: # secondary neutrons that
go on to produce subsequent fission

$k < 1$ subcritical

$k = 1$ critical (self-sustaining)

$k > 1$ supercritical

$$P(t) = P_0 k^{\frac{t}{\tau_{\text{gen}}}}$$

or 3) enrichment above natural abundance

(use neutron absorbers to keep reaction from going supercritical)
eg Cadmium control rods

[Richard Rhodes, Making of the Atomic Bomb]

Just talk about this:

fr-13

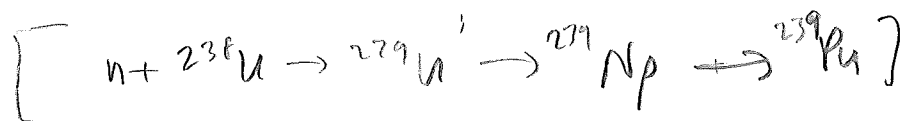
[What if want a supercritical rxn, ie a bomb?]

- slow n fission not suitable, energy release causes U to expand & $\therefore$ n's will all escape
- fast n fission of ^{238}U impossible because $k < 1$.
- fast n fission of ^{235}U is possible

enrichment: electromagnetic separator } Oak Ridge, TN
gaseous diffusion } (\$2 Billion)

Little Boy ~ 80% ^{235}U (about 50 kg)

(critical mass)
~10cm



^{239}Pu is fissionable

[Fat Man]

Hanford, WA

Nuclear physics

Mentor, biography (Sime)
(also Philip Ball: *seeing the forest*)

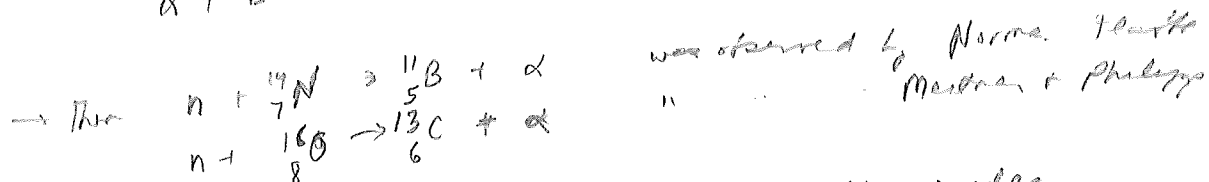
1919 Rutherford induced nuclear rxns $\gamma\alpha$



but not until 1924 was it realized that α was absorbed, not scattered.
(Blackett). The rxns $\Rightarrow$ existence of a strong attractive force that could overcome Coulomb repulsion

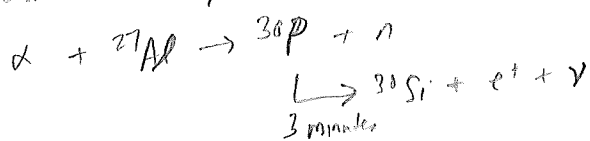
Nuclear fission: K now absorbed α 's, however, now did $\text{Li}, \text{Be}, \text{C}, \dots$

$\rightarrow$ ~~But~~ (1930) Rutherford & Bothe discovered ^{a new penetrating type of} radiation for α (from Po) bombardment of $\text{Be}, \text{Li}, \text{B}, \dots$. Joliot & Irene Curie (1932) found it to radiate γ from paraffin, & interpreted it as p , and Chadwick proved it was really neutrons (1932)

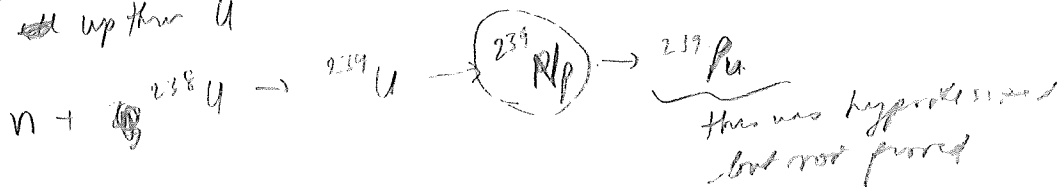


1934 Fermi ~~the~~ β decay, which used Pauli's ν idea

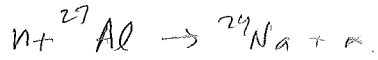
1934 Joliot & Curie produced artificial radioactivity



1934 Fermi ~~the~~ ~~the~~ & Rabi for bombarded with n
producing artificial radioactivity γ $\text{F}, \text{Al}, \text{Si}, \text{P}, \text{Cu}, \dots$
set up thru U



Fermi observed (n, α) rxns & (n, p) rxns on lighter elts



and (n, γ) rxns on heavy elements



and for the radioisotopes, which always emitted e^-

~~Mentore~~ + ~~others~~ decided to start studying transmutations.

Szilard + Chalmers observed



~~that~~ (lower energy) n could not be absorbed by Al, Na, Si
Mentore found that n could not be absorbed by Al, Na, Si
which required higher energies to speed off an α or p
but only in capture process by heavier nuclei Ag, I, Au

She suggested that slow neutrons are more easily captured
than fast neutrons

1934 Fermi showed that paraffin, etc slowed the n , after which by
were more easily absorbed by Ag, Cu, I, ~~etc~~

Uranium enrichment

Manh. project {
→ electromagnetic separation
→ gaseous diffusion

→ large ~~size~~ centrifuges (most common + economical)

3-5% ^{235}U ⇒ power reactors

70-90% ^{238}U ⇒ weapons

Hiroshima ⇒ 64 kg of ^{235}U ⇒ 15 kilotons
could do it w/ 25 kg (size of a melon)

Trinity ⇒ 6 kg of ^{239}Pu ⇒ 20 kilotons
could do it w/ 5 kg or less (size of a plum)