

2023
revision

$\beta-1$

[Next we turn to weak interaction + β -decay]

A dec. is an other transformation)

not generally occur provided it

does not violate any conservation laws

["what is not forbidden is mandatory"]

[review of p. 4] Conservation laws

• energy

• momentum

• angular momentum

a consequence of invariance
of laws of physics under

translations in space and time, + rotations in space
(Noether's theorem)

• electric charge

a consequence of gauge invariance
of electromagnetism

(believed to be absolute)

• baryon number
(quark number)

• lepton number

(no known symmetry)
"accidental" but valid for standard model

due to no known symmetry

but satisfied by the 4 known fermions;

quarks + leptons don't convert into each other;

If a process obeys all the conservation laws, work.

It is said to be "kinematically allowed"

otherwise "kinematically forbidden"

Energy conservation

[types of energy]

• rest energy mc^2

• kinetic energy T .

$$E = \gamma mc^2 \quad \text{where } \gamma = \frac{1}{\sqrt{1 - (\frac{v}{c})^2}} = 1 + \frac{1}{2} \frac{v^2}{c^2} + \frac{3}{8} \frac{v^4}{c^4} + \dots$$

$$T = E - mc^2 = (\gamma - 1)mc^2 \quad \text{exact}$$

$$\text{non-rel. approx.} \approx \underbrace{\left[\frac{1}{2}mv^2 \right]}_{\text{approx}} + \frac{3}{8}m \frac{v^4}{c^2} + \dots$$

• potential energy V

associated w/ interactions between particles (Em, strong...)

[Well before and well after the process the particles are well separated
 $\Rightarrow$ ignore any potential energies in initial & final states]

$$E_{\text{init}} = E_{\text{final}}$$

$$\sum_{\text{init}} (mc^2 + T) = \sum_{\text{final}} (mc^2 + T)$$

Neither kinetic energy nor mass is separately conserved
only their sum.

Define the kinetic energy released in a process

$$Q = T_{\text{final}} - T_{\text{init}}$$

$$= \sum_{\text{init}} mc^2 - \sum_{\text{final}} mc^2$$

If $Q = 0$, "elastic process" $\Rightarrow$ kinetic energy is conserved
[e.g. scattering of particles which retain their identities]

If $Q < 0$, "inelastic process" $\Rightarrow$ kinetic energy is lost
final state has more mass than initial (collisions)

[lumps of putty: hot object has slightly more mass than a cool one]

[More massive particles can be produced in a collision
e.g. cosmic rays, particle accelerators]

If $Q > 0$, "superelastic process"
 $\Rightarrow$ mass "converted" to kinetic energy
(e.g. fission, decay)

If initial state consists of a single particle m_0 at rest ($T_{\text{init}} = 0$)
then $Q = T_{\text{final}} \geq 0$

$$\Rightarrow m_0 c^2 \geq \sum_{\text{final}} mc^2$$

$\Rightarrow$ massive particles can only decay into less massive particles
(not vice versa)

Generally, decays w/ larger Q
have larger probability to occur.

+ therefore shorter half-lives

though for differing reasons

e.g. α -decay, higher $Q \rightarrow$ small potential barrier
to tunnel through

β -decay, high $Q \rightarrow$ large phase space

But in some cases, other considerations
cause the relationship to be violated

e.g. parity violation of the weak force
means that $\pi \rightarrow \mu$ is more likely
than $\pi \rightarrow e$,

despite the smaller phase space available

34

Conservation laws guarantee stability of certain particles

e^- is stable because it is least massive p.c. w/ electric charge

[if it decayed into lighter neutral particles, violates charge cons.]

Why is p stable?

[not the lightest positively charged p.c.: $p \rightarrow e^+ \gamma$]

1938 ^{Ernst} Stueckelberg proposed conservation of baryon # (A)
(quark #)

p is stable because it is lightest p.c. w/ baryon #

All forces of standard model conserve baryon # and lepton #
but extensions of standard model (e.g. GUTs)
typically violate both

Proton decay would be a signal of new (BSM) physics

Proton decay exps since 1970's
(Super Kamiokande)
Null results so far

[Problem]

$$\tau_p > 2 \times 10^{29} \text{ yrs}$$

[need a lot of protons]

(2023)

Why Chris asked about lifetime limits & confidence levels quoted in PPB

prob decay τ limit

B4.1

Decay rate $R = \frac{1}{\tau}$

Prob of decay in time of experiment T : $p = RT$

N nuclei

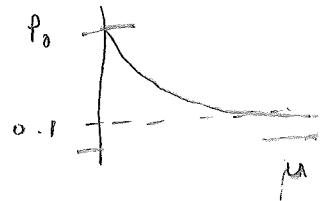
mean # nuclei decays in time T is $\mu = Np = \frac{NT}{\tau}$

Poisson distrib. P_n : prob of n nuclei decay

$$= e^{-\mu} \frac{\mu^n}{n!}$$

Prob. of no nuclei decays $P_0 = e^{-\mu}$

Suppose no nuclei decay.



This is very likely for small μ .

Unlikely for large μ

$\rightarrow$ large τ

$\rightarrow$ small τ

Set $e^{-\mu} = 0.1$

$$e^{\mu} = 10$$

$$\mu = \log(10) = 2.3$$

by 10% confidence we can say that $\mu \leq 2.3$

$\rightarrow$ we only have a 10% chance that no particles would decay in time T

$$\frac{NT}{\tau} < \ln 10$$

$$\tau > \frac{NT}{\ln 10}$$

Mass/energy conversions

Masses of atoms or nuclei are specified in terms of unified atomic mass unit (u).

$$1 \text{ u} \equiv \frac{\langle \text{mass of neutral } ^{12}\text{C atom in ground state} \rangle}{12}$$

[amu was based on ^{16}O ; u is based on ^{12}C]

A mole of particles = Avogadro's number of particles

$$N_A = 6.022\,140\,76 \times 10^{23} \text{ (exact)} \quad \text{[cf P1B]}$$

A mole of particles of mass 1u has mass of 1g (approx.)

omit (used to be a definition, & N_A was determined experimentally. Now N_A is defined exactly, so the statement is only approximate. Molar mass of ^{12}C is 11.999 999 9958(36) g.)

$$1 \text{ u} = \frac{10^{-3} \text{ kg}}{N_A} = 1.660\,539 \dots \times 10^{-27} \text{ kg (approx.)}$$

Using exact $c = 299\,792\,458 \frac{\text{m}}{\text{s}}$, a particle of mass 1u has rest energy

$$mc^2 = 1.492\,418 \dots \times 10^{-10} \text{ J (approx.)}$$

(exact)

$$c = 299792458 \frac{\text{m}}{\text{s}}$$

$$h = 6.62607015 \times 10^{-34} \text{ J/Hz}$$

$$e = 1.602176634 \times 10^{-19} \text{ C}$$

$$k_B = 1.380649 \times 10^{-23} \text{ J/K}$$

do not discuss

1 second defined by

 $\nu = 9,192,631,770$ cycles of hyperfine transition of ^{133}Cs .

$$E = h\nu = 6.091102297 \dots \times 10^{-24} \text{ J}$$

$$\text{so } 1 \text{ J} = 1.64 \times 10^{23} \text{ photons of } ^{133}\text{Cs}$$

$$E = \frac{h\nu}{e} = 3.8017170261 \text{ eV}$$

$$\text{ie } 1 \text{ eV} = 26,300 \text{ photons}$$

$$m = \frac{h\nu}{c^2} = 1.0136 \dots \times 10^{-42} \text{ kg}$$

= mass equivalent of photon

This defines the kg.

• We'll use eV rather than J

$$e = 1.602176 [634] \times 10^{-19} \text{ C}$$

$$1 \text{ eV} = 1.602176 [634] \times 10^{-19} \text{ J}$$

(exact) omit

(exact)

[recall $J = C \cdot V$]

A particle of mass 1u has rest energy

$$mc^2 = 931.494 \times 10^6 \text{ eV}$$

$$= 931.494 \text{ MeV}$$

(approx.)

(approx.)

omit

$$1 \text{ u} \Leftrightarrow 931.494 \dots \text{ MeV}$$

need the precision to talk about mass differences

$$\text{proton } [1.0072765 \dots] = 938.272 \text{ MeV}$$

$$\text{neutron } [1.008664 \dots] = 939.565 \text{ MeV}$$

$$\text{electron } [0.0005486 \dots] = 0.511 \text{ MeV}$$

[about $\frac{1}{1836} m_p$]

Neutron decay

$$n \rightarrow p$$

[baryon # conserved
but violates charge cons.]

$$n \rightarrow p e^-$$

[violates lepton #]

$$n \rightarrow p e^- \bar{\nu}_e$$

[actually lepton # was not known until 1950's
energy conservation led Pauli to predict ν .
We'll discuss later, but meanwhile want to be correct
so we'll include it, even]

$$m_{\bar{\nu}} \lesssim 1 \text{ eV}$$

[actually need to check if kinematically allowed]

$$Q = m_n c^2 - m_p c^2 - m_e c^2 - m_{\bar{\nu}} c^2$$

$$= 0.782 \text{ MeV}, \text{ kinematically allowed} \quad [\text{recall } \tau = 15 \text{ min.}]$$

A. nuclear process involving e^- or e^+ is called β -decay

[not ionization of atom]

most unstable light nuclei decay via β -decay

Mass excess

β-7

[It is generally easier to measure the mass of neutral atoms than of bare nuclei.]

A neutral atom has Z protons, N neutrons, Z electrons,

has an approximate mass $m \approx (Z+N) u = A u$

ie rest energy

$$mc^2 \approx A (931.494 \text{ MeV})$$

It is convenient to characterize the discrepancy as

$$mc^2 \equiv A (931.494 \text{ MeV}) + \Delta$$

← atomic mass
(incl. electrons)

$$\Delta = \text{"mass excess"} \quad [\text{actually rest energy excess}]$$

$$m = \left(A + \frac{\Delta}{931.494} \right) u$$

[nuclear mass only]

Recall: The unified mass unit u is defined so that $\Delta \equiv 0$ for $^{12}_6\text{C}$

[Using our earlier numbers]

$$p = 938.272$$

$$n = 939.565$$

B-8

• proton $\Delta = 6.778 \text{ MeV}$

• neutron $\Delta = 8.071 \text{ MeV}$

[mention agree w/ nuclear wallet cards, but not proton] [why not??]

• hydrogen = proton + electron $\Delta = 7.289 \text{ MeV}$ ✓

$$n \rightarrow p + e^- + \bar{\nu}_e$$

From now on, we can omit the factor of c:

$$Q = m_n - m_p - m_e - m_{\bar{\nu}_e}$$

$$= m_n - (m_H - m_e) - m_e$$

$$= m_n - m_H$$

$$= (1 \text{ u} + \Delta(n)) - (1 \text{ u} + \Delta(^1\text{H}))$$

$$= \Delta(n) - \Delta(^1\text{H})$$

$$= 0.782 \text{ MeV}$$
 ✓

Drop the c.

(if this freedom comes responsibility)

(need restore units when doing calculations)

Bound states

β-9

Deuteron $D = p n$

[Simplest composite nucleus
p and n bound together by strong force
(residue of color force between quarks)]

Composite objects held together due to their binding energy.

Binding energy = difference between the
sum of (rest) energies of the components
and the (rest) energy of the composite object

Deuteron binding energy:

$$B_D = m_n + m_p - m_D$$

$$= m_n + [m(^1\text{H}) - m_e] - [m(^2\text{H}) - m_e]$$

$$= [1u + \Delta(n)] + [1u + \Delta(^1\text{H})] - [2u + \Delta(^2\text{H})]$$

$$= \Delta(n) + \Delta(^1\text{H}) - \Delta(^2\text{H})$$

[NB. $B \neq \Delta$]

$$= 8.071 + 7.289 - 13.136$$

$$= 2.224 \text{ MeV} \quad (\text{about } 0.1\% \text{ of } m_d \sim 2 \text{ GeV})$$

$$\Rightarrow m_D = m_n + m_p - B_D$$

Deuteron mass is less than its constituents

[How to understand B ?]

Binding energy is due to attractive potential energy
minus kinetic energy

Consider a bound state of 2 particles
(eg hydrogen atom, or deuteron)

$$E_{12} = E_1 + E_2 + V_{12}$$

$\uparrow$ bound state energy $\uparrow$ energy of each particle $\uparrow$ (negative) potential energy due to attractive force

$$E_1 = m_1 + T_1$$

$$E_2 = m_2 + T_2$$

$$V_{12} = -|V_{12}|$$

Bound state at rest $E_{12} = m_{12}$

$$\Rightarrow m_{12} = m_1 + m_2 + \underbrace{(T_1 + T_2 - |V_{12}|)}_{-B_{12}}$$

$$\Rightarrow B_{12} = |V_{12}| - T_1 - T_2$$

Why do individual particles have kinetic energy?

Heisenberg uncertainty principle due to confinement to small space

$$\Delta p \Delta x \geq \frac{\hbar}{2}$$

$$\Delta p \geq \frac{\hbar}{2\Delta x} \quad \text{where } \Delta x \sim \text{size of bound state}$$

$$\langle p^2 \rangle - \langle p \rangle^2 = (\Delta p)^2$$

$$\text{Nonrelativistic: } \left. \begin{aligned} p &= mv \\ T &= \frac{1}{2}mv^2 \end{aligned} \right\} \Rightarrow T = \frac{p^2}{2m}$$

$$T \sim \frac{(\Delta p)^2}{2m} \sim \frac{\hbar^2}{2m(\Delta x)^2}$$

electron in hydrogen atom: $\Delta x \approx a_0$ (Bohr radius)

$$T = \frac{\hbar^2}{2m_0 a_0^2} = 13.6 \text{ eV}$$

$$|V_{12}| = 2T \quad (\text{virial theorem})$$

$$\Rightarrow \underline{B = 13.6 \text{ eV}} \quad (\text{binding energy of hydrogen})$$

deuteron: $\Delta x \approx 1 \text{ fm}$

$$T_p \approx T_n \sim \frac{\hbar^2}{2m(\Delta x)^2} = \frac{(\hbar c)^2}{2mc^2(\Delta x)^2} = \frac{(200 \text{ MeV fm})^2}{2(1000 \text{ MeV})(1 \text{ fm})^2} = \underline{\underline{20 \text{ MeV}}}$$

$$\text{Since } B_D = |V_{pn}| - T_p - T_n \approx 2 \text{ MeV}$$

$$\text{we infer } |V_{pn}| \approx 42 \text{ MeV}$$

(old notes)

~~β₁₀~~

First, let's understand binding energy for hydrogen atom.

$$E_H = (m_p + T_p) + (m_e + T_e) + V_{ep}$$

↑
Coulomb potential energy
between p and e

$$T_e = \frac{p^2}{2m_e}, \quad T_p = \frac{p^2}{2m_p} \ll T_e \quad \text{since } m_p \gg m_e$$

Estimate T_e . Uncertainty principle $\Rightarrow p \sim \frac{\hbar}{\Delta x}$

where Δx = uncertainty in position of electron $\sim a_0$ = Bohr radius

$$T_e \approx \frac{1}{2m_e} \left(\frac{\hbar}{a_0} \right)^2 = 13.6 \text{ eV}$$

Attractive inverse square force

$$\Rightarrow V_{ep} = -2T_e \quad (\text{virial theorem})$$

$$\Rightarrow T_e + T_p + V_{ep} \approx T_e + 0 - 2T_e = -T_e = -13.6 \text{ eV}$$

$$E_H = m_p + m_e + T_p + T_e + V_{ep} = m_p + m_e - 13.6 \text{ eV}$$

But $m_H = m_p + m_e - B_{\text{atomic}}$ so $B_{\text{atomic}} = 13.6 \text{ eV}$

$$\approx 1 \text{ GeV} + \frac{1}{2} \text{ MeV} - 10 \text{ eV}$$

↑

B is only 10^{-8} of total mass

Binding energy of deuteron

$$m_d = m_p + m_n + \underbrace{T_p + T_n + V_{pn}}_{-B_d}$$

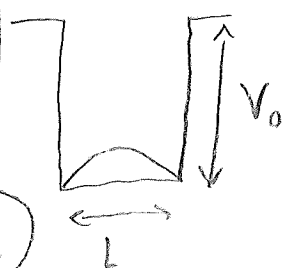
strong interaction
potential energy

where $B_d \approx 2.2 \text{ MeV}$

[Unlike Coulomb force, we have] No simple formula for V_{pn}

[nor do we expect one, since p and n are composites]

Use simplistic potential well model to estimate kinetic energy



$$T = T_p + T_n = 2 \cdot \frac{p^2}{2m} = 2 \cdot \frac{3}{2m} \left(\frac{h}{\lambda} \right)^2$$

$\lambda = \text{de Broglie wavelength}$

$$\lambda = 2L$$

$$T = \frac{3}{m} \left(\frac{2\pi\hbar}{2L} \right)^2 = \frac{3\pi^2}{m} \left(\frac{\hbar}{L} \right)^2$$

$$= \frac{3\pi^2}{(1000 \text{ MeV})} \left(\frac{200 \text{ MeV} \cdot \text{fm}}{L} \right)^2 = 1200 \text{ MeV} \left(\frac{\text{fm}}{L} \right)^2$$

$L = 2R$, $R = 2^{1/3} r_0 \approx 1.5 \text{ fm} \Rightarrow L \sim 3 \text{ fm}$

$$T \approx 130 \text{ MeV}$$

This is a very crude estimate (probably too high by factor of 2 or 4)

but suggests $V_0 \approx -T \approx -130 \text{ MeV}$

Simpler: $T = T_p + T_n \approx \frac{p^2}{m} \approx \frac{1}{m} \left(\frac{\hbar}{\Delta x} \right)^2 = \frac{1}{m} \left(\frac{\hbar c}{\Delta x} \right)^2 = \frac{1}{1000 \text{ MeV}} \left(\frac{200 \text{ MeV} \cdot \text{fm}}{1 \text{ fm}} \right)^2 \approx 40 \text{ MeV}$

Suggests $V \approx -42 \text{ MeV} \Rightarrow B = -T - V \approx 2 \text{ MeV}$

old note

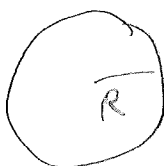
~~B-11.1~~

particle in 3d box



$$T = 3 \cdot \frac{1}{2m} \left(\frac{h}{\lambda} \right) = \frac{3}{2m} \left(\frac{2\pi\hbar}{2L} \right)^2 = \frac{3\pi^2\hbar^2}{2mL^2}$$

particle in a sphere



$$-\frac{\hbar^2}{2m} \frac{1}{r} \frac{d^2}{dr^2} (r\psi) = E\psi$$

$$\frac{d^2}{dr^2} (r\psi) + \frac{2mE}{\hbar^2} (r\psi) = 0$$

$$\psi = \frac{\sin kr}{r} \quad \text{where} \quad \frac{\hbar^2 k^2}{2m} = E$$

$$\psi(R)=0 \Rightarrow kR=\pi \Rightarrow k=\frac{\pi}{R} \Rightarrow \boxed{E = \frac{\hbar^2 \pi^2}{2mR^2}}$$

If $L = \underbrace{2R}_{\text{diameter}}$ then 3d box $\Rightarrow \frac{3\pi^2\hbar^2}{8mR^2}$, slightly less $\uparrow$

If volume equal $L^3 = \frac{4\pi}{3} R^3$

$$\text{the 3d box} = \frac{3\pi^2\hbar^2}{2m} \left(\frac{3}{4\pi} \right)^{2/3} \frac{1}{R^2} = \underline{\underline{0.577 \frac{\hbar^2 \pi^2}{mR^2}}}$$

Slightly more (by 15%)

[Because $B_d > 0$, deuteron can't fall apart]

$D \rightarrow p + n$ kinematically forbidden $[Q = -B_d < 0]$

[but what about]

$D \xrightarrow{?} pp e^- \bar{\nu}_e$ [Can n inside D decay?]

$$\begin{aligned}
 Q &= m_D - 2m_p - m_e - \cancel{m_{\nu}}^0 \\
 &= (m(^2\text{H}) - m_e) - 2(m(^1\text{H}) - m_e) - m_e \\
 &= m(^2\text{H}) - 2m(^1\text{H}) \\
 &= [2u + \Delta(^2\text{H})] - 2[1u + \Delta(^1\text{H})] \\
 &= \Delta(^2\text{H}) - 2\Delta(^1\text{H}) \\
 &= -1.442 \text{ MeV} \quad \text{still kinematically forbidden}
 \end{aligned}$$

Free neutron is unstable but

neutron bound to proton is stable

[due to strong force which reduces its energy]

Tritium ${}^3\text{H}$ (pnn)



$$Q = [m({}^3\text{H}) - m_e] - [m({}^3\text{He}) - 2m_e] - m_e - m_\nu$$

$$= [3u + \Delta({}^3\text{H})] - [3u + \Delta({}^3\text{He})] - m_\nu$$

$$= \Delta({}^3\text{H}) - \Delta({}^3\text{He}) - m_\nu$$

$$= 0.019 \text{ MeV} - m_\nu \quad \text{kinematically allowed (barely)}$$

$$\hookrightarrow T_{1/2} \sim 12 \text{ years}$$

$$\begin{array}{l} m({}^3\text{H}) = 2809.4496 \text{ MeV} \\ m({}^3\text{He}) = 2809.4310 \text{ MeV} \\ \hline 0.0186 \text{ MeV} \end{array}$$

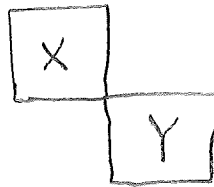
• tritium important in setting upper bound on ν mass

eg: $m_\nu \leq 19 \text{ keV}$ for decay to occur

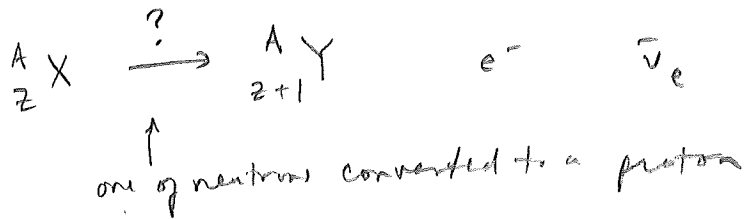
$$Q = T_e + \cancel{T_{\text{H}}} + T_\nu$$

↓
0

by measuring $\langle T_e \rangle_{\text{max}}$
can put more stringent bound on m_ν

β^- decay

β^- decay always connects isobars
[same # baryons]



[baryon #, lepton #, charge]

$$\begin{aligned}
 Q &= m_{\text{nuc}}(X) - m_{\text{nuc}}(Y) - m_e \\
 &= [A u + \Delta(X) - Z m_e] - [A u + \Delta(Y) - (Z+1) m_e] - m_e \\
 &= \Delta(X) - \Delta(Y)
 \end{aligned}$$

If $\Delta(X) > \Delta(Y)$, then X is unstable to β^- decay

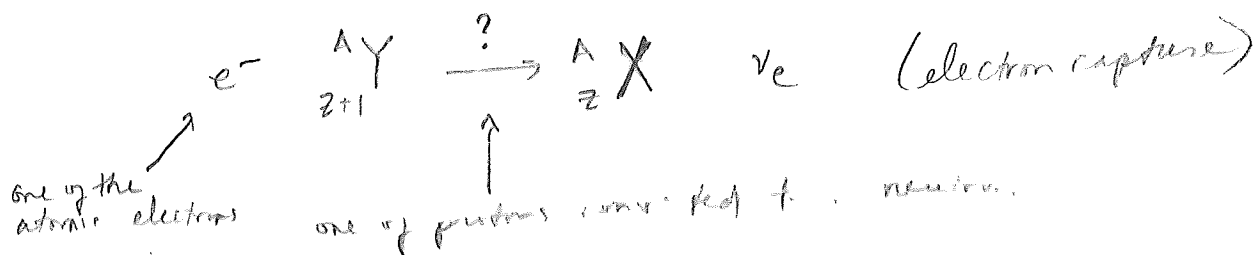
What if $\Delta(X) < \Delta(Y)$?

Can a proton convert to a neutron?



kinematically
forbidden for e^-, p at rest
(but formation of a neutron star)

[conservation laws obeyed]



$$Q = m_e + m_{\text{nuc}}(Y) - m_{\text{nuc}}(X)$$

$$= m_e + [Au + \Delta(Y) - (Z+1)m_e] - [Au + \Delta(X) - Zm_e]$$

$$= \Delta(Y) - \Delta(X)$$

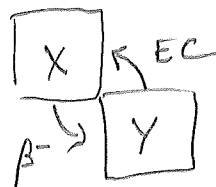
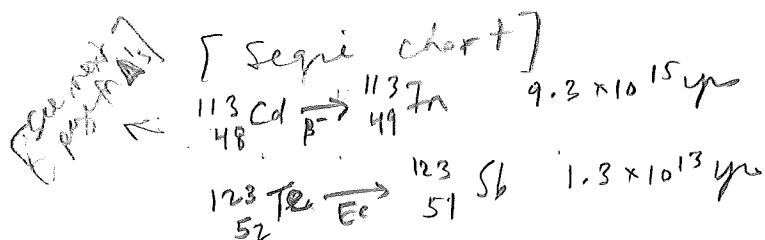
If $\Delta(Y) > \Delta(X)$, the Y is unstable to EC (electron capture)

EC is followed by X-ray emission, as one of outer shell electrons fills the vacancy created by the s-shell electron

[first observed 1938 Alvarez : of Krane, p. 272]

Given two adjacent isobars, one will always be unstable.

(but next to adjacent isobars can exist)



(line shows as stable in Segre chart).



$$\Delta = -89.051$$

$$T_{1/2} = 9.3 \times 10^5 \text{ yrs}$$

β -decay
12.27%

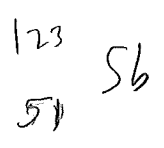
$$J = \frac{1}{2} +$$



$$\Delta = -89.367$$

$$4.3\%$$

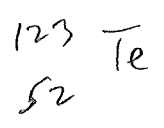
$$J = \frac{9}{2} +$$



$$\Delta = -89.223$$

$$42.7\%$$

$$J = \frac{7}{2} +$$



$$\Delta = -89.172$$

$$1.3 \times 10^{13} \text{ yrs (EC)}$$

0.902%

$$J = \frac{1}{2} +$$

β^+ decay

$\beta-16$

[baryon #, lept-#, charge]



↓
proton converted to neutron

$$Q = m_{\text{nuc}}(Y) - m_{\text{nuc}}(X) - m_e$$

$$= [A u + \Delta(Y) - (z+1)m_e] - [A u + \Delta(X) - z m_e] - m_e$$

$$= \Delta(Y) - \Delta(X) - 2m_e$$

$$\text{If } \Delta(Y) - \Delta(X) > 2m_e = 1.022 \text{ MeV}$$

then Y is unstable to β^+ decay (also $E\gamma$)

β^+ decay " followed by



$$Q = m_{e^+} + m_{e^-} = 1.022 \text{ MeV}$$

Each γ carries $\sim 0.511 \text{ MeV}$ (full-line sign of e^+e^- annihilation)

[first observed 1934 (Johst-Curie; of Kross p.272)]

Recall neutral atom

$$m(X) = A u + \underbrace{\Delta(X)}_{\text{mass excess}} \quad \text{experimental}$$

931.494 MeV

Also

$$m(X) = Z m_p + Z m_e + N m_n - \underbrace{B}_{\substack{\text{binding energy of nucleus} \\ (+ \text{ atom}) \\ \text{negligible}}}$$

$$= Z m(^1\text{H}) + N m_n - B$$

($\uparrow$ ignore 13.6 eV)

$$= Z (1u + \Delta(^1\text{H})) + N (1u + \Delta(n)) - B$$

$$= A u + Z \Delta(^1\text{H}) + N \Delta(n) - B$$

Compare

$$B = Z \underbrace{\Delta(^1\text{H})}_{7.289} + N \underbrace{\Delta(n)}_{8.071} - \Delta(X)$$

[check: B for ^1H and n]

Rewrite in terms of A and N-Z

$$B = \underbrace{(N+Z)}_A (7.680 \text{ MeV}) + (N-Z) (0.391 \text{ MeV}) - \Delta(X)$$

For ^{12}C , $A=12$, $N=Z$, $\Delta=0$ so

$$B(^{12}\text{C}) = 12 (7.680 \text{ MeV})$$

For nuclei similar to ^{12}C , $N \approx Z$ and $\Delta \approx 0$

$$B = (7.7 \text{ MeV}) A$$

i.e. binding energy per nucleon is approx const.

[see curve of binding energy: fig 3.16]

$$m(X) = Zm(^1\text{H}) + Nm_n - B$$

$$\approx A (939 \text{ MeV}) - A (8 \text{ MeV})$$

$$\approx (931 \text{ MeV}) A$$

Nuclei are $\sim 1\%$ less massive than sum of constituents

[deuteron only 0.1% $\Rightarrow$ weak binding]
 $\downarrow$
see curve!

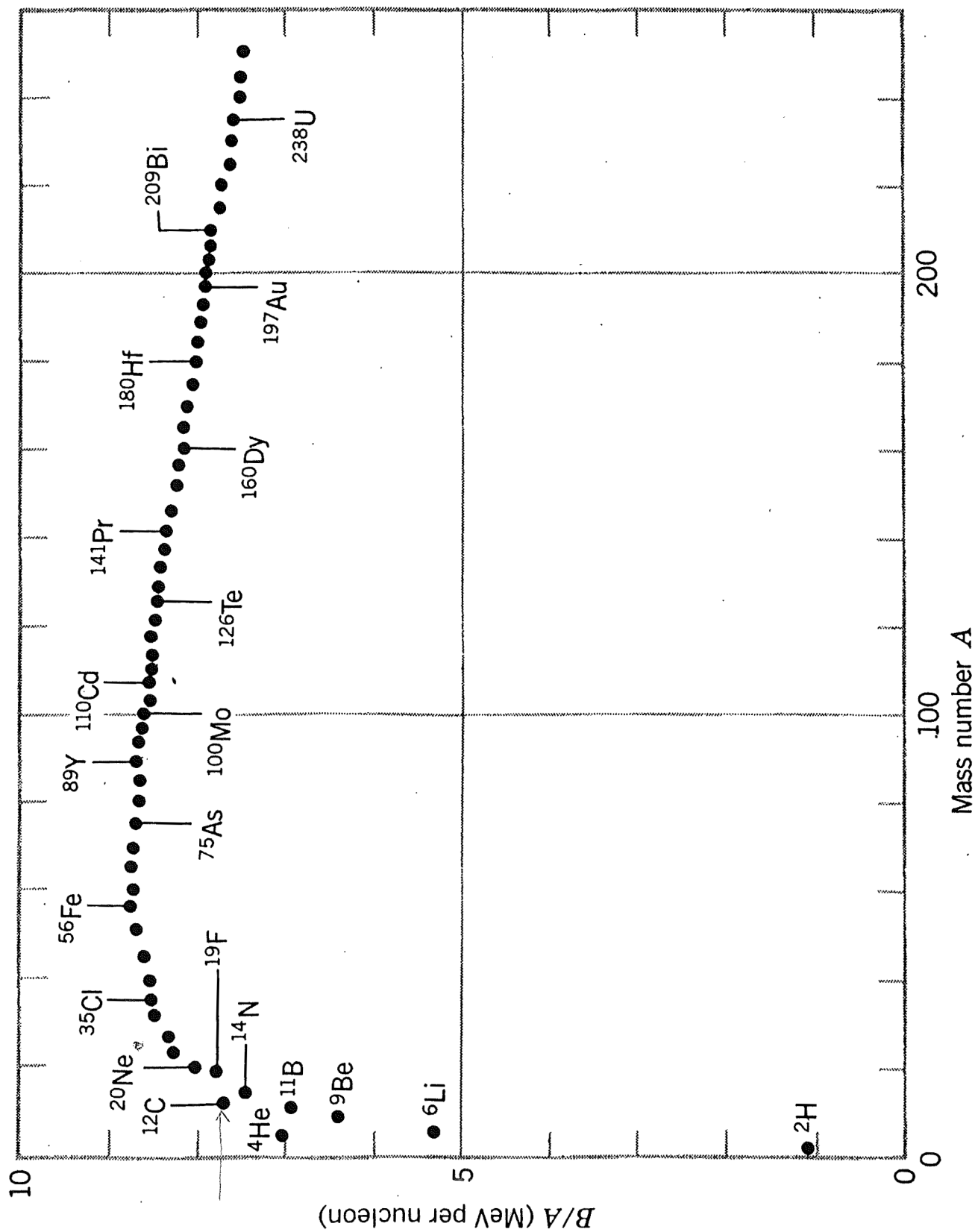


Figure 3.16 The binding energy per nucleon.