

Nuclear transmutation

Rutherford & collaborators used α particles emitted by Po & Ra
 typical $T = 5 \text{ MeV}$ to probe atomic structure.

For heavy targets (Z large), Coulomb barrier prevents
 α from getting too close to nucleus,
 which therefore behaves as a point charge

$$\frac{d\sigma}{d\Omega} \sim \frac{1}{\sin^4 \frac{\theta}{2}}$$

Scattering from lighter nuclei begins to depart from this
 suggesting a finite size for nucleus.

If α gets close enough ($\sim 1 \text{ fm}$), strong force
 can cause some of its particles to be absorbed

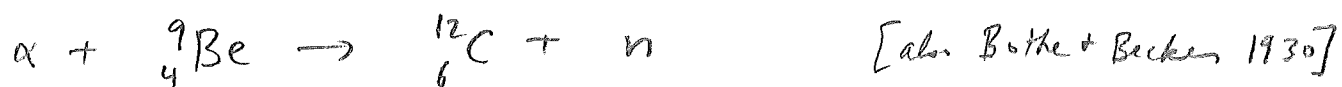
Nuclear transmutation (Rutherford, 1919)



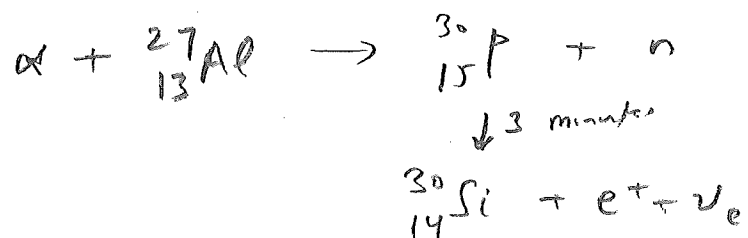
Cockcroft/Walton [designed a linear accelerator to
 accelerated protons to $\sim 0.5 \text{ MeV}$, and
 [with this, they] "split the atom" (1932)]



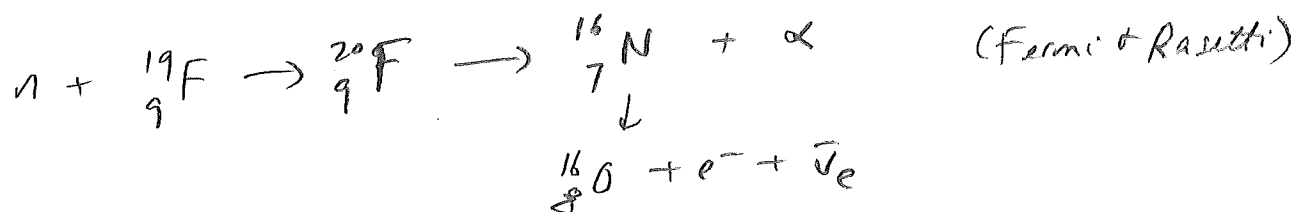
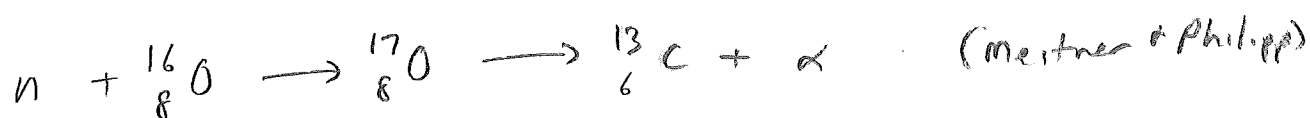
Chadwick discovers the neutron (1932)



Artificially radioactive isotopes (Joliot + I. Curie, 1934)

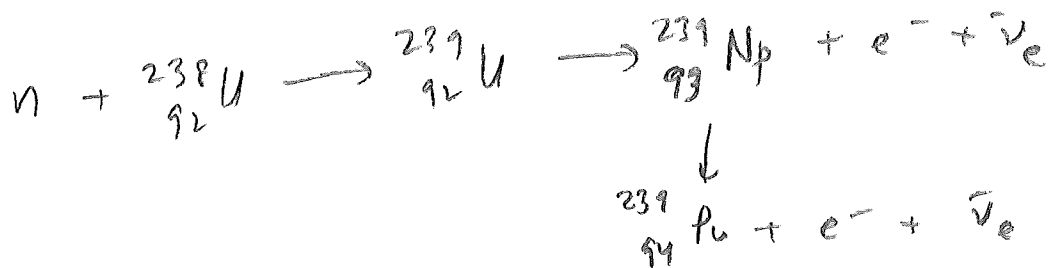


Neutrons as probes [no Coulomb barrier; but can't see inside them either]



Many artificially radioactive isotopes produced
[nothing below fluorine]

even beyond uranium (transuranic)





[It was found that] for medium energy neutrons ($T = 20 \text{ MeV}$) the absorption cross section for neutrons is

$$\sigma \sim \pi R_x^2$$

Suggests most neutrons are absorbed by an nucleus they strike

Recall $R_x \sim A^{1/3} r_0$ where $r_0 = 1.2 \text{ fm}$

$$\sigma \sim A^{2/3} \pi r_0^2$$

For heavy nuclei: $\pi A^{2/3} \sim 100$

[$A \sim 180$; tungsten]

$$\sigma \sim (10 \text{ fm})^2 \sim 1 \text{ barn} \dots$$

Nuclear physicists define

$$1 \text{ barn} = (10 \text{ fm})^2 = 100 \text{ fm}^2 = 10^{-28} \text{ m}^2$$

"broadside of a barn"

Uranium ($A = 238$)

Uranium $\sim 1.7 \text{ barns}$

$$\sigma \sim (13 \text{ fm})^2$$

(4.1)

For higher energies, $\sigma \downarrow \Rightarrow$ not all neutrons are absorbed
"partial transparency"

For slow neutrons, $\sigma \uparrow$

[cf graph from Evans]

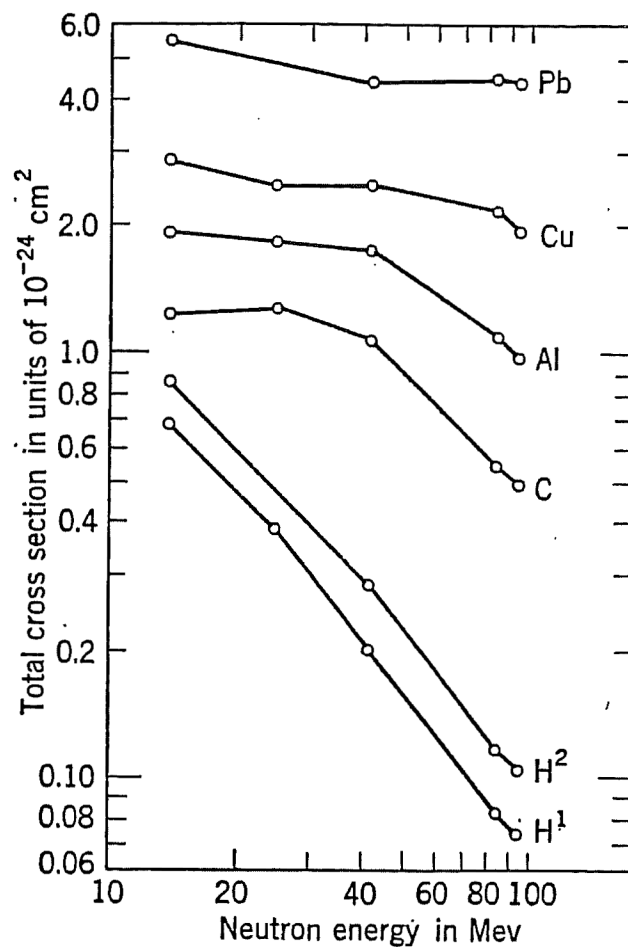
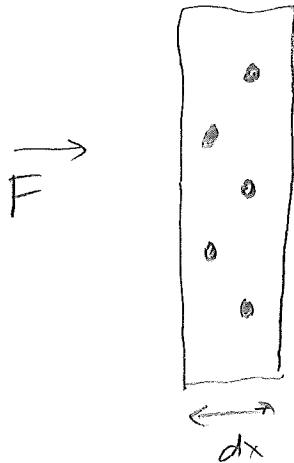


Fig. 2.3 Total neutron cross sections σ_t of five elements for neutrons of 14, 25, 42, 84, and 95 Mev. The decrease in σ_t at the higher neutron energies suggests that all neutrons which strike the target area $\pi(R + \lambda)^2$ are not captured and that nuclear matter is slightly transparent to swift neutrons. [From Hildebrand and Leith (H51).]

Let incident flux of neutrons be F



Probability of absorption by nuclei in film of thickness dx

$$f = n \sigma (dx)$$

n = # density of nuclei

$n(A dx)$ = # targets, area of targets: $n \sigma A dx$

$\Rightarrow$ Flux correspondingly diminishes

$$dF = -(n \sigma dx) F$$

$$\frac{dF}{F} = -n \sigma dx$$

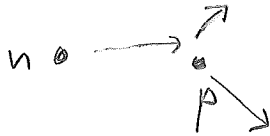
$$\ln F = -n \sigma x + \ln F_0$$

$$F = F_0 e^{-n \sigma x} = F_0 e^{-\frac{x}{l}}$$

$\boxed{l = \frac{1}{n \sigma} = \text{mean free path of neutrons}}$

$$\left[\begin{array}{l} n = \frac{\rho}{A} \cdot N_A \quad \rho = \text{mass density} \\ U \rightarrow \rho = 19 \frac{\text{g}}{\text{cm}^3} \Rightarrow n = 4.8 \times 10^{28} \text{ m}^{-3} \\ \sigma = 1.7 \times 10^{-28} \text{ m}^2 \\ l \approx 0.12 \text{ m} \end{array} \right] \leftarrow \text{[HW]}$$

Neutrons, being neutral, cannot be accelerated or slowed
 but they can be slowed down by collisions, or by
 absorption + re-emission



Fermi found that absorption $\sigma \uparrow$ for lower energy n

[see graph for michel]

Cadmium ~ 2000 barns for slow neutrons

[empirically $\sigma \sim \frac{1}{v}$]

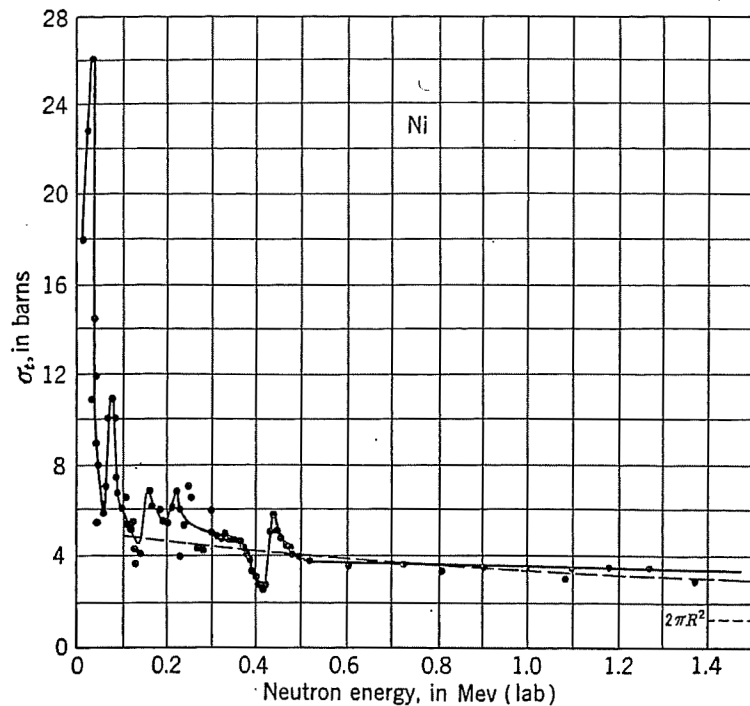


Fig. 2.1 The total cross section (absorption plus scattering) of nickel for neutrons from 0.01 to 1.5 Mev. Approximately monoenergetic neutrons (± 20 kev below 0.5 Mev; ± 150 kev above 0.5 Mev) were produced by bombarding a thin Li target with protons accelerated by an electrostatic generator. The dotted line is from the continuum theory of Feshbach and Weisskopf (F49), assuming a nuclear radius of 4.6×10^{-13} cm. [The experimental data are from Barschall, Bockelman, and Seagondollar (B15).]

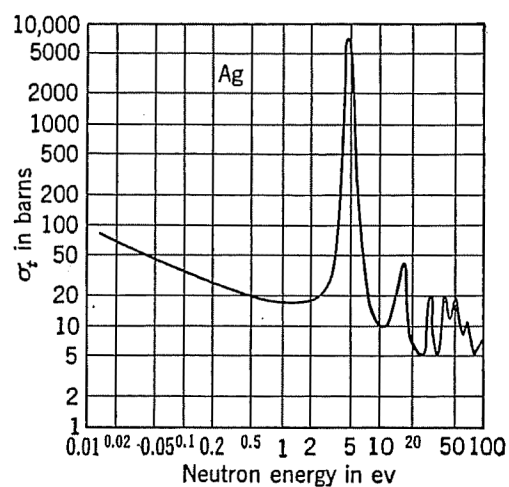
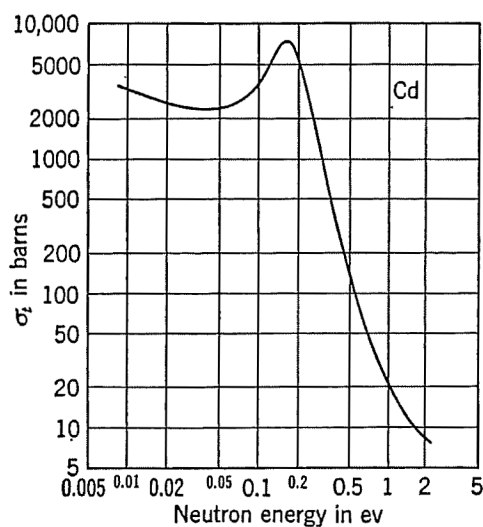
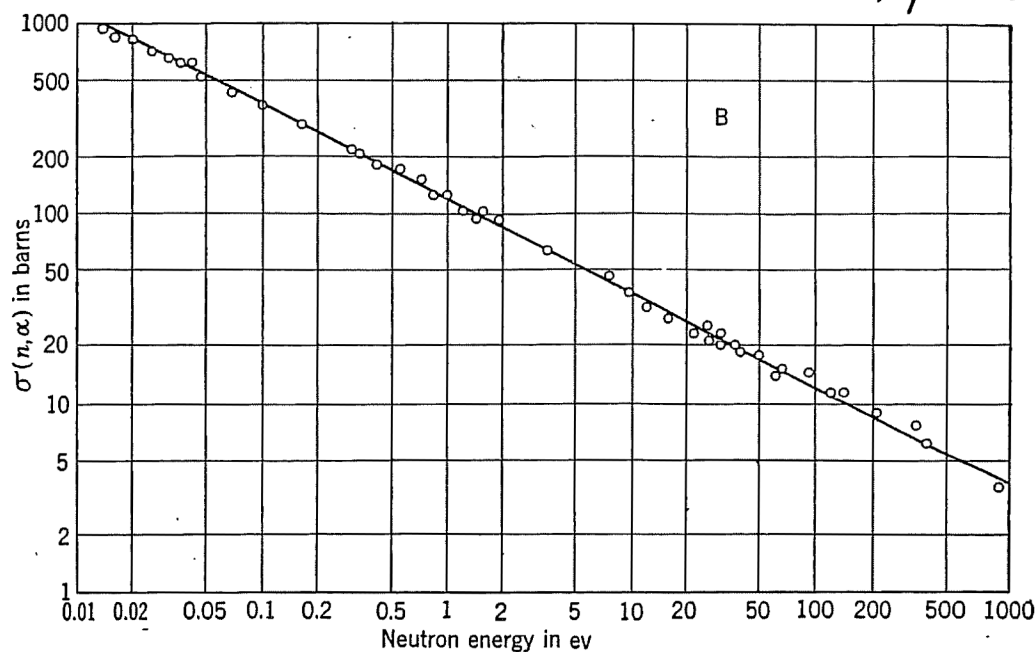


Fig. Introduction.1 Total neutron cross sections for B, Cd, and Ag, illustrating typical combinations of $1/V$ dependence (pure, in the case of boron) and slow-neutron resonances. The resonance in Cd at $E_0 = 0.176$ eV has a total width $\Gamma = 0.115$ eV and a maximum value of $\sigma_0 = 7,200$ barns. The curves are for the normal isotopic mixtures. In Cd, the 0.176-eV resonance is due entirely to Cd^{113} (relative abundance, 12.3 per cent); hence $\sigma_0 \simeq 57,000$ barns for Cd^{113} alone. The complex γ -ray spectrum leading to the ground level of Cd^{114} has been measured by Bartholomew and Kinsey (B16). [From Goldsmith, Ibser, and Feld (G29) and Hughes (H68).]

[skipped in 2019]

fr. 6

Why? wave properties of neutrons

de Broglie wavelength $\lambda = \frac{h}{p}$

Define reduced wavelength $\tilde{\lambda} = \frac{\lambda}{2\pi} = \frac{h}{p}$

Consider non-relativistic neutrons ($T \ll m_n c^2 \sim 1000 \text{ MeV}$)

$$T = \frac{p^2}{2m} = \frac{1}{2m} \left(\frac{h}{\tilde{\lambda}} \right)^2 = \frac{1}{2mc^2} \left(\frac{hc}{\tilde{\lambda}} \right)^2 = \frac{1}{2(1000 \text{ MeV})} \left(\frac{200 \text{ MeV} \cdot \text{fm}}{\tilde{\lambda}} \right)^2$$
$$= (20 \text{ MeV}) \left(\frac{1 \text{ fm}}{\tilde{\lambda}} \right)^2$$

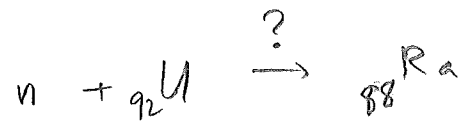
$$\frac{\tilde{\lambda}}{1 \text{ fm}} = \sqrt{\frac{20 \text{ MeV}}{T}}$$

For $T \gtrsim 20 \text{ MeV}$, $\tilde{\lambda} < 1 \text{ fm}$, neutron acts as a particle
($\tilde{\lambda} \ll R$)

Small T , $\tilde{\lambda} \gg R$, acts like a wave, $\sigma \propto \phi$

Neutrons w/ $T \sim \frac{3}{2} kT \sim 0.05 \text{ eV}$ are called "thermal neutrons"

1938 Hahn & Strassmann (chemists)
thought they observed



but no physical process could explain this

- why did they think it was radium?
- Because it had similar chemical properties to Ba and a radioactive product was precipitating out of soln $\rightarrow$ Ba
- But they then tried to separate Ra from Ba and failed
- What did they do? Consulted a physicist.

Former colleague Lise Meitner

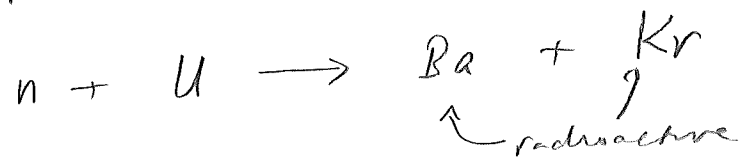
an Austrian Jew.

When Hitler annexed Austria, she lost her citizenship

In 1938, she escaped to Sweden.

She discussed it w/ nephew, Otto Frisch
and decided it was Ba.

1939 Meitner & Frisch propose



"nuclear fission"

Why didn't anyone think of this before?

$$\frac{B}{A} \text{ for U} \sim 7.5 \text{ MeV}$$

$$\frac{B}{A} \text{ for Ba, Kr} \sim 8.2 \text{ MeV}$$

about 0.7 MeV released per nucleon

$\Rightarrow \sim 170 \text{ MeV per fission.}$

$\rightarrow$ kinetic energy of decay products

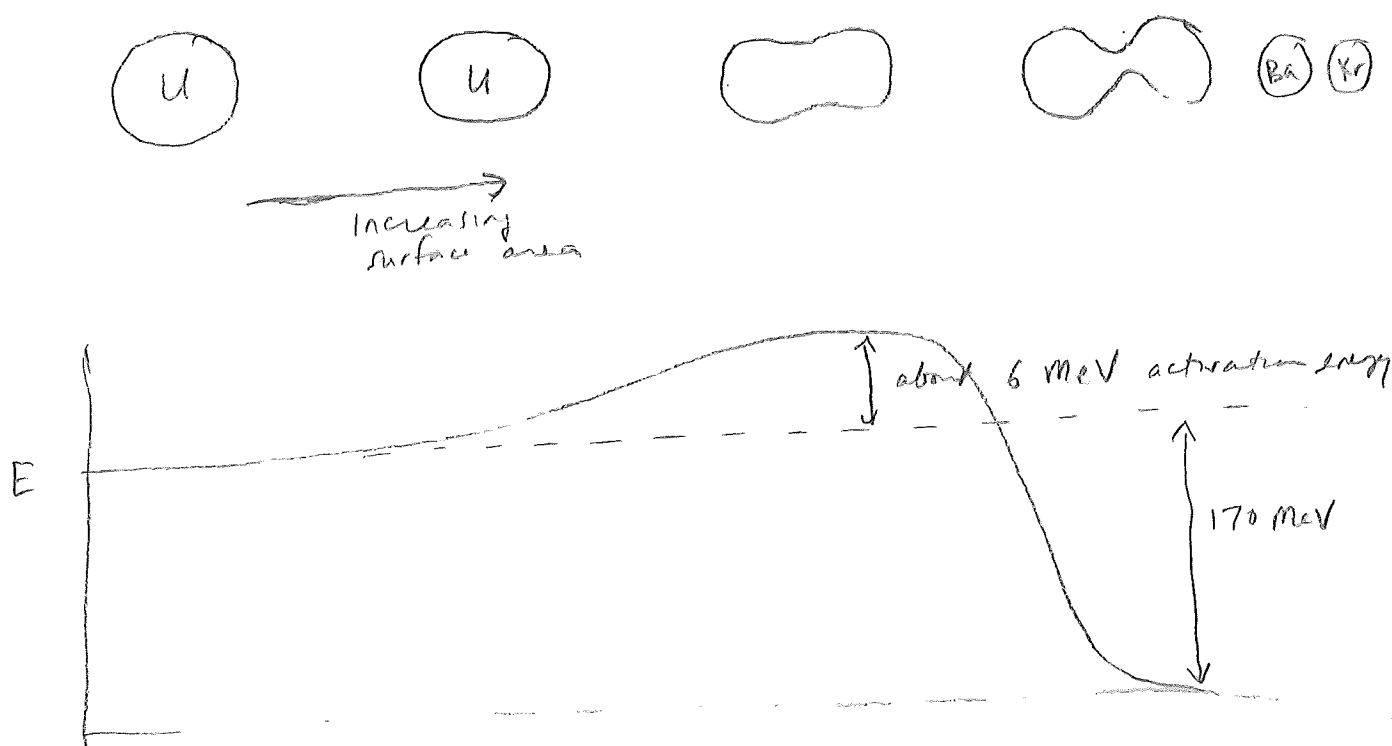
Frick in his lab verified energy of fragments.

fi-9

If U less stable than Ba, Kr why doesn't it just decay?

Think α -decay. Coulomb barrier holds them in.

Alternative picture. Liquid drop model (surface tension)



Tunnelling through barrier can occur but is rare
"spontaneous fission"

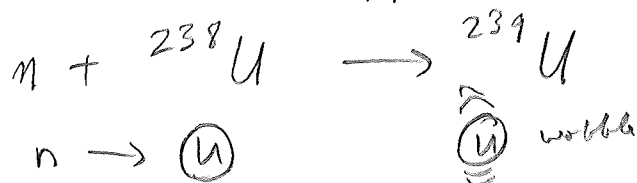
$$T_{\frac{1}{2}} \sim 3.5 \times 10^{17} \text{ years}$$

[HKR, p. 1169]

(α -decay is more likely)

Incident neutron supplies energy

fi-10



$$Q = \Delta({}^{238}\text{U}) + \Delta(n) - \Delta({}^{239}\text{U}) = 4.8 \text{ MeV}$$

$$\begin{bmatrix} 47.306 \\ 8.071 \\ -51.571 \\ \hline 4.806 \end{bmatrix}$$

energy released causes drop to wobble

but activation energy of ${}^{239}\text{U}$ is 6.6 MeV

[? 5.7 MeV in
HRK, 1156]

neutrons must have $E > 1.8 \text{ MeV}$
to initiate fission

"fast neutron fission of ${}^{238}\text{U}$ ", $\frac{v}{c} \sim \sqrt{\frac{2T}{mc^2}} \sim 0.06$

But Hahn & Strassmann were using slow (thermal) neutrons
($T < 1 \text{ eV}$)

1939 Bohr realized that they were observing
"slow neutron fission of ${}^{235}\text{U}$ "

${}^{238}\text{U}$ 99.3%
 ${}^{235}\text{U}$ 0.7%



$$Q = 6.5 \text{ MeV}$$

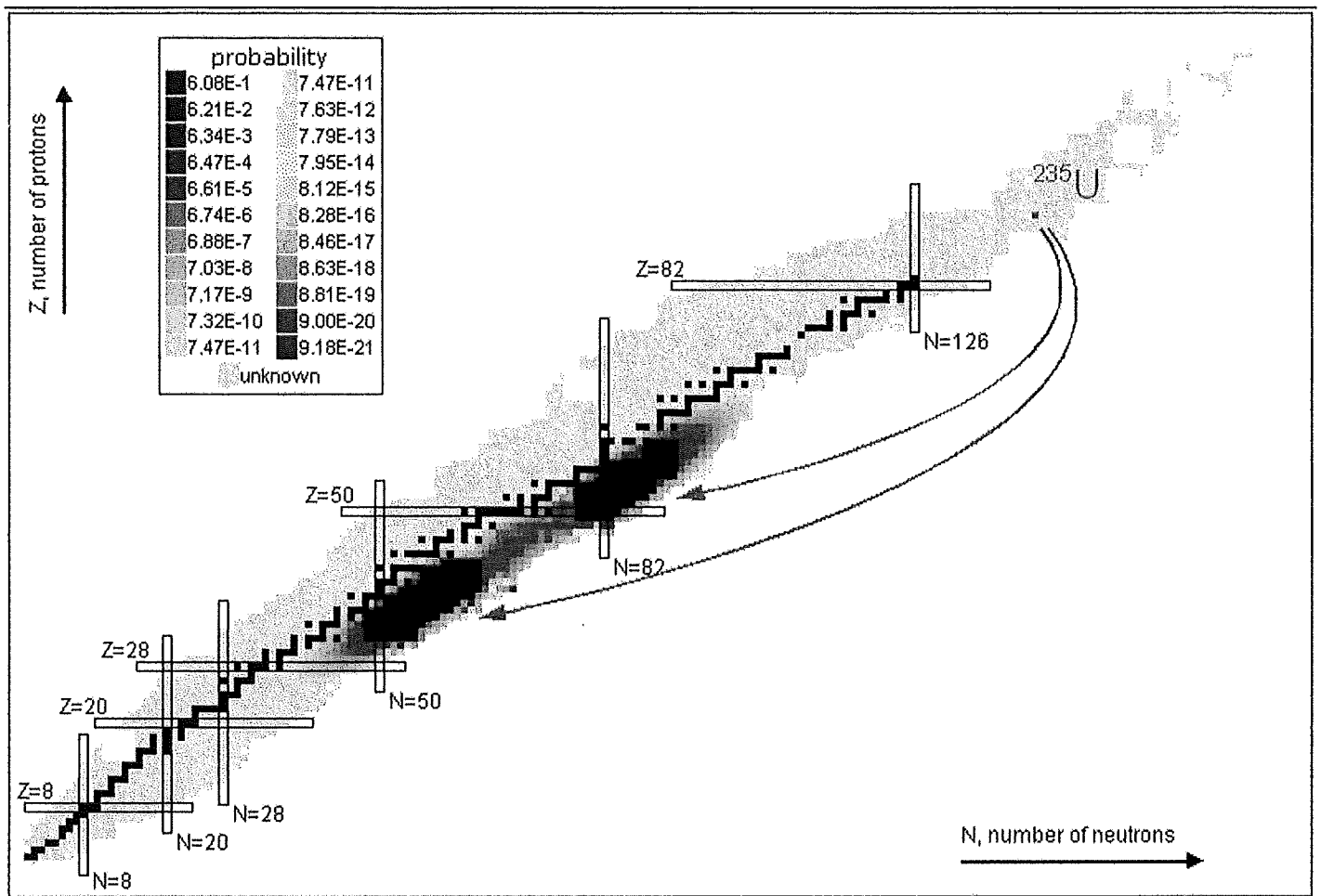
$\Rightarrow$ why difference? even $\rightarrow$ odd
odd $\rightarrow$ even

and activation energy of ${}^{236}\text{U}$ is 6.2 MeV

no additional kinetic energy required!

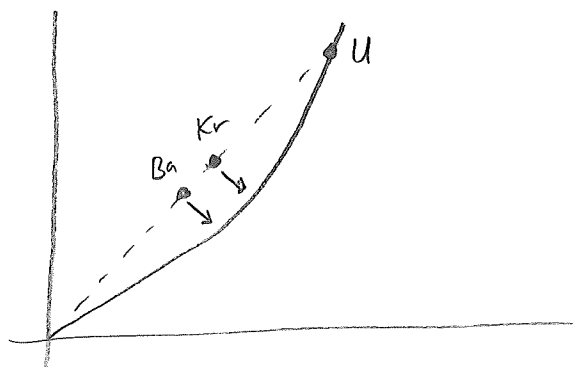
$$\begin{bmatrix} 40.916 \\ 8.071 \\ -42.442 \\ \hline 6.545 \end{bmatrix}$$

[? 5.2 in HRK]



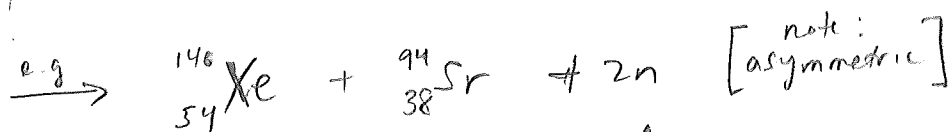
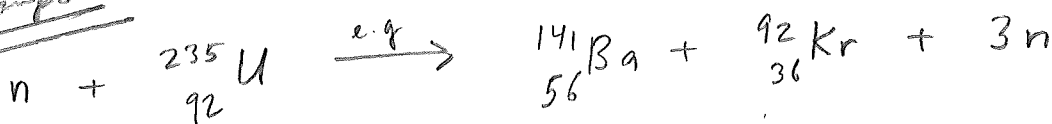
Lab Deming Lounge ~~Hyper~~
blogspot.com

[Why are Ba + Kr radioactive? Because they are neutron rich]



Fission often releases 2 or 3 neutrons in addition to fission fragments

For example



Released neutrons can initiate further fissions
 $\Rightarrow$ chain reaction

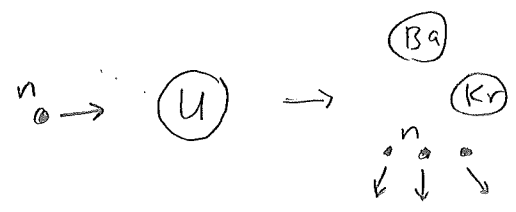
controlled $\Rightarrow$ nuclear reactor
 uncontrolled $\Rightarrow$ weapons

about 11.6 MeV released per fission

vs 1-10 eV released in chemical reactions $\Rightarrow$ factor of 10^7 to 10^8

Leo Szilard convinced Einstein to write to Roosevelt...

First controlled nuclear reaction (Dec 2, 1942)
 U of Chicago squash court (Fermi)



Secondary neutrons
 avg $H \sim 2.47$ / fission
 avg $T \sim 2$ MeV
 (some are delayed)

escape through surface
 w/o causing fission

↓ solution
 increase volume to
 surface ratio
 (critical mass)
 ↓ depends on
 material for pelt

For sustained rxn, need

- 1) critical mass
- 2) pure moderator

But k = multiplication factor = # secondary neutrons that
 go on to produce subsequent fission

- $k < 1$ subcritical
- $k = 1$ critical (self-sustaining)
- $k > 1$ supercritical

$$P(t) = P_0 k^{\frac{t}{\tau_{\text{gen}}}}$$

or 3) enrichment above natural abundance
 (use neutron absorbers to keep reaction from going supercritical)
 e.g. Cadmium control rods

[Richard Rhodes, Making of the Atomic Bomb]

Just talk about this:

fr-13

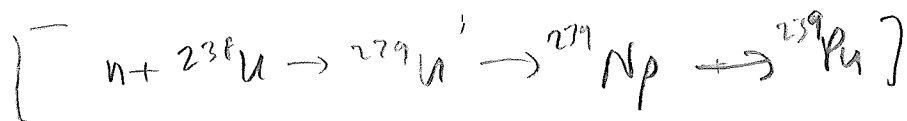
[What if want a supercritical rxn, ie a bomb?]

- slow n fission not suitable, energy release causes U to expand & $\therefore$ n's will all escape
- fast n fission of ^{238}U impossible because $k < 1$.
- fast n fission of ^{235}U is possible

enrichment: electromagnetic separation } Oak Ridge, TN
gaseous diffusion } (\$2 Billion)

Little Boy ~ 80% ^{235}U (about 50 kg)

(critical mass
~ 100 kg)



^{239}Pu is fissionable

[Fat Man]

Hanford, WA

Nuclear physics

Methods: biography (Sime)

(also Philip Ball:
serving the Rock)

1979 Ruthenium induced nuclear rxns γ &

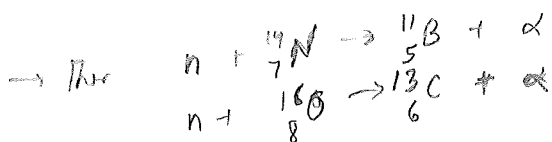


but not until 1924 was it realised that α was absorbed, not scattered.
(Blackett). The α rays $\Rightarrow$ existence of a strong attractive force that
overcame Coulomb repulsion.

Nuclei to emit β near absorbed α 's, however, nor did
1i, Re, C, ?.

→ Burr (1930) Kall & Burke discovered ^{a new penetrable type of} ~~radiation~~ ^{for α (for P_0) bombardment}

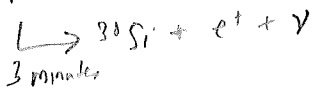
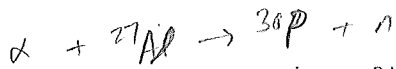
of Be, Li, B. . Joliot & Irene Curie (1932) found it to radiate β particles
P from paraffin, & interpreted it as β , but Chadwick proved
it was really neutrons (1932)



was observed by Norm. F. H.
" " " " Marston & Phillips

1934 Form. Re. J. B. S. which used Pauli's γ idea

1934 Tolstoy & Lamm produced artificial radioactivity



1934 Farm ~~8~~ ¹⁰⁰ + Records included etc 7 n
 2 1 A

producing extremely radioactive 2. Al, Si, P, Cu. '11
 still up there U



This was hypothesized
but not proved

Fermi observed (n, α) and (n, p) reactions on lighter elements



and (n, γ) reactions



and if the intermediate nucleus always emitted e^-

Mentzer + Hahn decided to start studying transmutation.

Szilard + Chalmers observed



~~that~~ (lower energy) n could not be absorbed by Al, Na, Si which required higher energies to split off α or p s but only in capture processes by heavier nuclei Ag, I, Au

She suggested that slow neutrons are more easily captured than fast neutrons

1934 Fermi showed that paraffin, etc slowed the n , after which they were more easily absorbed by Ag, Cu, I, etc.

Uranium enrichment

meth.
 process {
 → electromagnetic separation
 → gaseous diffusion

→ large ~~gas~~ centrifuges (most common + economical)

3-5% ^{235}U ⇒ power reactors

70-90% ^{235}U ⇒ weapons

Hiroshima ⇒ 64 kg of ^{235}U ⇒ 15 kilotons
could do it w/ 25 kg (size of a melon)

Trinity ⇒ 6 kg of ^{239}Pu ⇒ 20 kilotons
could do it w/ 5 kg or less (size of a plum)