

2023
revision

$\beta-1$

A decay (or other transformation)

will generally occur provided it

does not violate any conservation laws

["what is not forbidden is mandatory"]

[review of p. 4] Conservation laws

- energy
- momentum
- angular momentum

} a consequence of invariance
of laws of physics under
translations in space and time, + rotations in space
(Noether's theorem)

↑ [optional]

- electric charge

} a consequence of gauge invariance
of electromagnetism

[believed to be absolute]

- baryon number
(quark number)
- lepton number

} "accidental" but valid for standard model
[due to no known sym. - bry
but satisfied by the 4 known forces;
quarks + leptons don't convert into each other;

If a process obeys energy + momentum conservation,

it is said to be "kinematically allowed"

otherwise "kinematically forbidden"

Energy conservation

[types of energy]

• rest energy mc^2 • kinetic energy T

$$E = \gamma mc^2 \quad \text{where } \gamma = \frac{1}{\sqrt{1 - (\frac{v}{c})^2}} = 1 + \frac{1}{2} \frac{v^2}{c^2} + \frac{3}{8} \frac{v^4}{c^4} + \dots$$

$$T = E - mc^2 = (\gamma - 1)mc^2$$

$$\text{non rel. approx.} \approx \frac{1}{2}mv^2 + \frac{3}{8}m \frac{v^4}{c^2} + \dots$$

• potential energy V

associated w/ interactions between particles (EM, strong...)

Well before and well after the process the particles are well separated
 $\Rightarrow$ ignore any potential energies in initial & final states

$$E_{\text{init}} = E_{\text{final}}$$

$$\sum_{\text{init}} (mc^2 + T) = \sum_{\text{final}} (mc^2 + T)$$

Neither kinetic energy nor mass is separately conserved
 only their sum.

Define the kinetic energy released in a process

$$Q = T_{\text{final}} - T_{\text{init}}$$

$$= \sum_{\text{init}} mc^2 - \sum_{\text{final}} mc^2$$

If $Q = 0$, "elastic process" $\Rightarrow$ kinetic energy is conserved
[e.g. scattering of particles which retain their identities]

If $Q < 0$, "inelastic process" $\Rightarrow$ kinetic energy is lost
final state has more mass than initial (collisions)

[lumps of putty: hot object has slightly more mass than a cool one]

[More massive particles can be produced in a collision
e.g. cosmic rays, particle accelerators]

If $Q > 0$, "superelastic process"
 $\Rightarrow$ mass "converted" to kinetic energy
(e.g. fission, decay)

If initial state consists of a single particle m_0 at rest ($T_0 = 0$)
then $Q = T_{\text{final}} \geq 0$

$$\Rightarrow m_0 c^2 \geq \sum_{\text{final}} mc^2$$

$\Rightarrow$ massive particles can only decay into less massive particles
(not vice versa)

Generally, decays w/ larger Q

have larger probability to occur

+ therefore shorter half lives

though for differing reasons

e.g. α -decay, higher Q $\rightarrow$ small potential barrier
to tunnel through.

β -decay, high Q $\rightarrow$ large phase space

But in some cases, other considerations

cause the relationship to be violated

e.g. parity violation of the weak force

means that $\pi \rightarrow \mu$ is more likely

than $\pi \rightarrow e$,

despite the smaller phase space available

Conservation laws guarantee stability of some particles

e^- is stable because it is least massive p.c. w/ electric charge

[if it decayed into lighter neutral particles, violates charge cons.]

Why is p stable?

[not the lightest positively charged p.c.: $p \rightarrow e^+ \gamma$]

1938 Stueckelberg proposed conservation of baryon # (A)
(quark #)

p is stable because it is lightest p.c. w/ baryon #

All forces of standard model conserve baryon # and lepton #
but extensions of standard model (e.g. GUTs)
typically violate both

Proton decay expts since 1970's
(Super Kamiokande)

[problem]

Null results so far

$$\tau_p > 2 \times 10^{29} \text{ yrs}$$

[need a lot of protons]

(2023)

Why Chris
asked about
lifetime limits & confidence levels
quoted in PPB

prob decay τ limits

B4.1

Decay rate $R = \frac{1}{\tau}$

Prob of decay in time of experiment T : $p = RT$

N nuclei

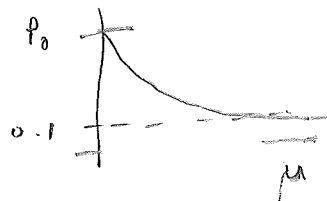
mean # nuclei decays in time T is $\mu \cdot Np = \frac{NT}{\tau}$

Poisson distrib. P_n : prob of n nuclei decaying

$$= e^{-\mu} \frac{\mu^n}{n!}$$

Prob. of no nuclei decaying $P_0 = e^{-\mu}$

Suppose no nuclei decay.



This is very likely for small μ .

Unlikely for large μ

$\rightarrow$ large τ

$\rightarrow$ small τ

Set $e^{-\mu} = 0.1$

$$e^{\mu} = 10$$

$$\mu = \ln(10) = 2.3$$

by 90% confidence we can say that $\mu \leq 2.3$

$\rightarrow$ we only have a 10% chance that no particles would decay in time T

$$\frac{NT}{\tau} < \ln 10$$

$$\tau > \frac{NT}{\ln 10}$$

Mass/energy conversions

Masses of atoms or nuclei are specified in terms of unified atomic mass unit (u).

$$1 \text{ u} \equiv \frac{\text{Mass of neutral } ^{12}\text{C atom in ground state}}{12}$$

[amu was based on ^{16}O ; u is based on ^{12}C]

A mole of particles = Avogadro's number of particles

$$N_A = 6.022\,140\,76 \times 10^{23} \text{ (exact)}$$

A mole of particles of mass 1u has mass of 1g (approx.)

[used to be a definition, & N_A was determined experimentally. Now N_A is defined exactly, so the statement is only approximate. Molar mass of ^{12}C is $11.999\,999\,995\,8(36) \text{ g}$.]

$$1 \text{ u} = \frac{10^{-3} \text{ kg}}{N_A} = 1.660\,539 \dots \times 10^{-27} \text{ kg (approx.)}$$

Using exact $c = 299\,792\,458 \frac{\text{m}}{\text{s}}$, a particle of mass 1u has rest energy

$$mc^2 = 1.492\,418 \dots \times 10^{-10} \text{ J (approx.)}$$

(exact)

$$c = 299792458 \frac{\text{m}}{\text{s}}$$

$$h = 6.62607015 \times 10^{-34} \text{ J/Hz}$$

$$e = 1.602176634 \times 10^{-19} \text{ C}$$

$$k_B = 1.380649 \times 10^{-23} \text{ J/K}$$

do not discuss

1 second defined by

 $\nu = 9,192,631,770$ cycles of hyperfine transition of ^{133}Cs .

$$E = h\nu = 6.091102297... \times 10^{-24} \text{ J}$$

$$\text{so } 1 \text{ J} = 1.64 \times 10^{23} \text{ photons of } ^{133}\text{Cs}$$

$$E = \frac{h\nu}{e} = 3.8017170261 \times 10^{-5} \text{ eV}$$

$$\text{ie } 1 \text{ eV} = 26,300 \text{ photons}$$

$$m = \frac{h\nu}{c^2} = 1.036... \times 10^{-42} \text{ kg}$$

= mass equivalent of photon

Thus defines the kg.

- We'll use eV rather than J

$$e = 1.602176634 \times 10^{-19} \text{ C} \quad (\text{exact})$$

$$1 \text{ eV} = 1.602176634 \times 10^{-19} \text{ J} \quad (\text{exact}) \quad [\text{recall } J = C \cdot V]$$

A particle of mass 1u has rest energy

$$mc^2 = 931.494 \times 10^6 \text{ eV} \quad (\text{approx.})$$

$$= 931.494 \text{ MeV} \quad (\text{approx.})$$

$$1 \text{ u} \Leftrightarrow 931.494... \text{ MeV}$$

[PPB]

$$\bullet \text{ proton } 1.0072765u = 938.272 \text{ MeV}$$

$$\bullet \text{ neutron } 1.0086649u = 939.565 \text{ MeV}$$

$$\bullet \text{ electron } 0.0005486u = 0.511 \text{ MeV}$$

} [keep these on the board]

[about $\frac{1}{1800} m_p$]

[Both about 1 GeV]

[Neutron slightly more massive, so decays]

$$n \rightarrow p$$

[violates charge cons.]

$$n \rightarrow p + e^-$$

[violates lepton number]

$$n \rightarrow p + e^- + \bar{\nu}_e$$

[still obey energy conservation?]

$$\bullet \text{ neutrinos } \lesssim 1 \text{ eV}$$

$$Q = m_n c^2 - m_p c^2 - m_e c^2 - m_{\bar{\nu}} c^2 = 0.782 \text{ MeV}$$

Neutron decay is kinematically allowed.

A nuclear process involving e^- or e^+ is called β -decayMost unstable light nuclei decay via β -decay

Mass excess

p-7

[It is generally easier to measure the mass of neutral atoms than of bare nuclei]

A neutral atom with Z protons, N neutrons, Z electrons

has an approximate mass $m \approx (Z+N) u = A u$

ie rest energy

$$mc^2 \approx A (931.494 \text{ MeV})$$

It is convenient to characterize the discrepancy as

$$mc^2 \equiv A (931.494 \text{ MeV}) + \Delta$$

← atomic mass
(incl. electrons)

$$\Delta = \text{"mass excess"} \quad [\text{actually rest energy excess}]$$

$$m = \left(A + \frac{\Delta}{931.494} \right) u$$

[nuclear vallet cords]

Recall: The unified ^{mass unit} u is defined so that $\Delta \equiv 0$ for $^{12}_6\text{C}$

[Using our earlier numbers]

β-8

• proton $\Delta = 6.778 \text{ MeV}$

• neutron $\Delta = 8.071 \text{ MeV}$

[mention agree w/ nuclear wallet cards, but not proton]

• hydrogen = proton + electron $\Delta = 7.289 \text{ MeV}$ ✓

$$n \rightarrow p + e^- + \bar{\nu}_e$$

From now on, we can omit the factor of c :

$$Q = m_n - m_p - m_e - m_{\bar{\nu}_e}$$

$$= m_n - (m_H - m_e) - m_{\bar{\nu}_e}$$

$$= m_n - m_H$$

$$= (1 \text{ u} + \Delta(n)) - (1 \text{ u} + \Delta(^1\text{H}))$$

$$= \Delta(n) - \Delta(^1\text{H})$$

$$= 0.782 \text{ MeV}$$
 ✓

Drop the c ,
(my free freedom
comes
responsibility)
need restore
units under
the calculation
...

Bound states

β-9

Deuteron $d = pn$

[Simplest composite nucleus
p and n bound together by strong force
(residue of color force between quarks)]

Composite objects held together due to their binding energy.

Binding energy = difference between the
sum of (rest) energies of the components
and the (rest) energy of the composite object

Deuteron binding energy:

$$\begin{aligned} B_d &= m_n + m_p - m_d \\ &= m_n + [m(^1\text{H}) - m_e] - [m(^2\text{H}) - m_e] \\ &= [1u + \Delta(n)] + [1u + \Delta(^1\text{H})] - [2u + \Delta(^2\text{H})] \\ &= \Delta(n) + \Delta(^1\text{H}) - \Delta(^2\text{H}) \end{aligned}$$

$$= 8.071 + 7.289 - 13.136$$

$$= 2.224 \text{ MeV} \quad (\text{about } 0.1\% \text{ of } m_d \sim 2 \text{ GeV})$$

$$\Rightarrow m_d = m_n + m_p - B_d$$

Deuteron mass is less than its constituents

[How to understand B?]

First, let's understand binding energy for hydrogen atom.

$$E_H = (m_p + T_p) + (m_e + T_e) + V_{ep}$$

$\uparrow$
 Coulomb potential energy
 between p and e

$$T_e = \frac{p^2}{2m_e}, \quad T_p = \frac{p^2}{2m_p} \ll T_e \quad \text{since } m_p \gg m_e$$

Estimate T_e . Uncertainty principle $\Rightarrow p \sim \frac{\hbar}{\Delta x}$

where Δx = uncertainty in position of electron $\sim a_0$ = Bohr radius

$$T_e \approx \frac{1}{2m_e} \left(\frac{\hbar}{a_0} \right)^2 = 13.6 \text{ eV}$$

Attractive inverse square force

$$\Rightarrow V_{ep} = -2T_e \quad (\text{virial theorem})$$

$$\Rightarrow T_e + T_p + V_{ep} \approx T_e + 0 - 2T_e = -T_e = -13.6 \text{ eV}$$

$$\text{But } m_H = m_p + m_e - B_{\text{atomic}} \text{ so } B_{\text{atomic}} = 13.6 \text{ eV}$$

$$\approx 1 \text{ GeV} + \frac{1}{2} \text{ MeV} - 10 \text{ eV}$$

$\uparrow$
 B is only 10^{-8} of total mass

more precisely

$$\left[\begin{aligned} T_e &= \frac{1}{2m} \left(\frac{h}{\lambda} \right)^2 = \frac{1}{2m} \left(\frac{\hbar}{a_0} \right)^2 \\ a_0 &= \frac{\hbar^2}{m k e^2} \\ T_e &= \frac{1}{2} m c^2 \left(\frac{k e^2}{\hbar c} \right)^2 = \frac{1}{2} m c^2 \alpha^2 \end{aligned} \right.$$

Binding energy of deuteron

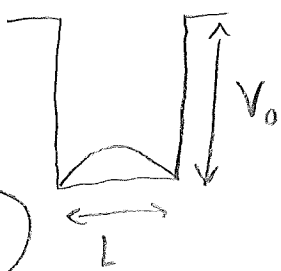
$$m_d = m_p + m_n + \underbrace{T_p + T_n + V_{pn}}_{-B_d}$$

strong interaction
potential energy

where $B_d \approx 2.2 \text{ MeV}$

[Unlike Coulomb force, we have] No simple formula for V_{pn}
[nor do we expect one, since p and n are composites]

Use simplistic potential well model to estimate kinetic energy



$$T = T_p + T_n = 2 \cdot \frac{p^2}{2m} = 2 \cdot \frac{3}{2m} \left(\frac{h}{\lambda} \right)^2$$

$\lambda = \text{de Broglie wavelength}$

$$\lambda = 2L$$

$$T = \frac{3}{m} \left(\frac{2\pi\hbar}{2L} \right)^2 = \frac{3\pi^2}{mc^2} \left(\frac{\hbar c}{L} \right)^2$$

$$= \frac{3\pi^2}{(1500 \text{ MeV})} \left(\frac{200 \text{ MeV} \cdot \text{fm}}{L} \right)^2 = 1200 \text{ MeV} \left(\frac{\text{fm}}{L} \right)^2$$

$$L = 2R, \quad R \cdot 2^{1/3} r_0 \approx 1.5 \text{ fm} \Rightarrow L \sim 3 \text{ fm}$$

$$T \approx 130 \text{ MeV}$$

This is a very crude estimate (probably too high by factor of 2 or 4)

$$\text{but suggests } V_0 \approx -T \approx -130 \text{ MeV}$$

Simpler: $T = T_p + T_n \approx \frac{p^2}{m} \approx \frac{1}{m} \left(\frac{\hbar}{\Delta x} \right)^2 = \frac{1}{mc^2} \left(\frac{\hbar c}{\Delta x} \right)^2 = \frac{1}{1000 \text{ MeV}} \left(\frac{200 \text{ MeV} \cdot \text{fm}}{1 \text{ fm}} \right)^2 \approx 40 \text{ MeV}$

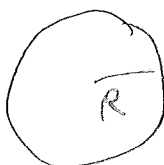
Suggests $V \approx -42 \text{ MeV} \Rightarrow B = -T - V \approx 2 \text{ MeV}$

particle in 3d box



$$T = 3 \cdot \frac{1}{2m} \left(\frac{h}{\lambda} \right)^2 = \frac{3}{2m} \left(\frac{2\pi\hbar}{2L} \right)^2 = \frac{3\pi^2\hbar^2}{2mL^2}$$

particle in a sphere



$$-\frac{\hbar^2}{2m} \frac{1}{r} \frac{d^2}{dr^2} (r\psi) = E\psi$$

$$\frac{d^2}{dr^2} (r\psi) + \frac{2mE}{\hbar^2} (r\psi) = 0$$

$$\psi = \frac{\sin kr}{r} \quad \text{where} \quad \frac{\hbar^2 k^2}{2m} = E$$

$$\psi(R)=0 \Rightarrow kR=\pi \Rightarrow k=\frac{\pi}{R} \Rightarrow \boxed{E = \frac{\hbar^2 \pi^2}{2mR^2}}$$

If $L = \underbrace{2R}_{\text{diameter}}$ then 3d box $\Rightarrow \frac{3\pi^2\hbar^2}{8mR^2}$, slightly less $\uparrow$

If volume equal $L^3 = \frac{4\pi}{3} R^3$

$$\text{the 3d box} = \frac{3\pi^2\hbar^2}{2m} \left(\frac{3}{4\pi} \right)^{2/3} \frac{1}{R^2} = \underline{\underline{0.577 \frac{\hbar^2 \pi^2}{mR^2}}}$$

slightly more (by 15%)

[deuteron is not unstable to $d \rightarrow p + n$, but what about β -decay?]

$$d \xrightarrow{?} p + p + e^- + \bar{\nu}_e$$

$$Q = m_d - 2m_p - m_e$$

$$= (m(^2\text{H}) - m_e) - 2(m(^1\text{H}) - m_e) - m_e$$

$$= m(^2\text{H}) - 2m(^1\text{H})$$

$$= \Delta(^2\text{H}) - 2\Delta(^1\text{H})$$

$$= 13.136 - 2(7.289)$$

$$= -1.442 \text{ MeV} \quad \text{Kinematically forbidden}$$

[Not as much difference as binding energy,
but still kinematically forbidden]

Free neutron is unstable, but neutron
bound to a proton is stable!

[strong force reduces the energy]

Tritium $t = pnn$

$$t \xrightarrow{?} d + p + e^- + \bar{\nu}_e$$

$$Q = \Delta(^3\text{H}) - \Delta(^2\text{H}) - \Delta(^1\text{H})$$

$$= 14.95 - 13.136 - 7.289$$

$$= -5.475 \text{ MeV} \quad (\text{forbidden})$$

[Not possible]

← probably omit this

[How does tritium decay?]

$$pnn \rightarrow p\bar{p}n + e^- + \bar{\nu}_e$$

$$Q = [m(^3\text{H}) - m_e] - [m(^3\text{He}) - 2m_e] - m_e$$

$$= [3u + \Delta(^3\text{H})] - [3u - \Delta(^3\text{He})]$$

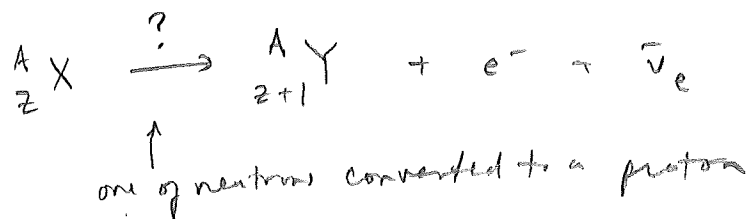
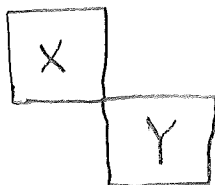
$$= \Delta(^3\text{H}) - \Delta(^3\text{He})$$

$$= 0.019 \text{ MeV} \quad (\text{allowed})$$

$$\begin{cases} m(^3\text{H}) = 2809.4496 \\ m(^3\text{He}) = 2809.4310 \\ \hline 0.0186 \end{cases}$$

just barely unstable

 β -decy always connects isobars = nuclei w/ same # nucleons

β^- decay

$$\begin{aligned}
 Q &= m_{\text{nuc}}(X) - m_{\text{nuc}}(Y) - m_e \\
 &= [A u + \Delta(X) - Z m_e] - [A u + \Delta(Y) - (Z+1) m_e] - m_e \\
 &= \Delta(X) - \Delta(Y)
 \end{aligned}$$

If $\Delta(X) > \Delta(Y)$, then X is unstable to β^- decay

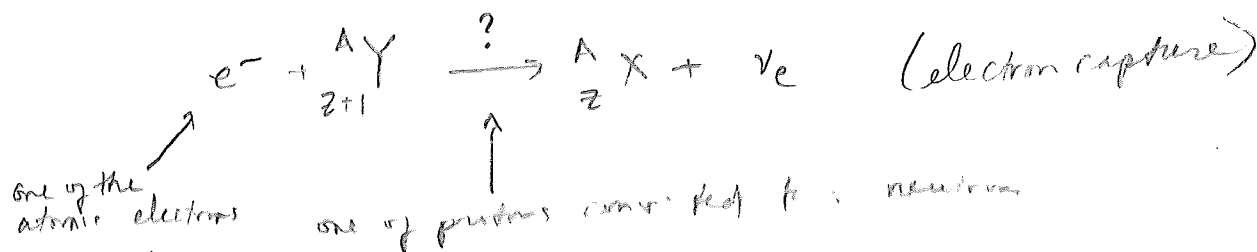
What if $\Delta(X) < \Delta(Y)$?

Can a proton convert to a neutron?



[conservation laws obeyed]

kinematically
forbidden for e^-, p at rest
(but formation of a neutron star)



$$\begin{aligned} Q &= m_e + m_{\text{nuc}}(Y) - m_{\text{nuc}}(X) \\ &= m_e + [A u + \Delta(Y) - (Z+1)m_e] - [A u + \Delta(X) - Z m_e] \\ &= \Delta(Y) - \Delta(X) \end{aligned}$$

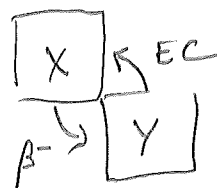
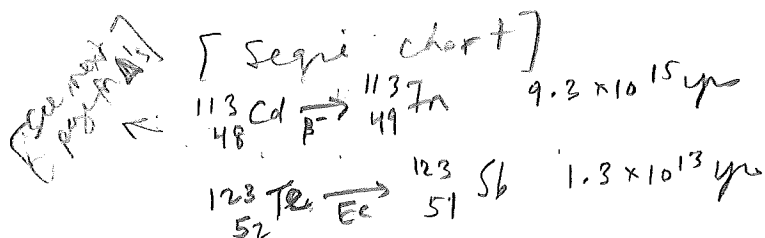
If $\Delta(Y) > \Delta(X)$, then Y is unstable to EC (electron capture)

EC is followed by X-ray emission, as one of outer shell electrons fills the vacancy created by the s-shell electron

[first observed 1938 Alvarez; cf Krane, p. 272]

Given two adjacent isobars, one will always be unstable.

(but next to adjacent isobars can exist)



(~~but~~ shows as stable in Segre chart)

$^{113}_{48}\text{Cd}$

$$\Delta = -89.051$$

$$T_{1/2} = 9.3 \times 10^5 \text{ yrs}$$

$$J = \frac{1}{2}^+$$

β -decay
12.22%



$^{113}_{49}\text{In}$

$$\Delta = -89.367$$

4.3%

$$J = \frac{9}{2}^+$$

$^{123}_{51}\text{Sb}$

$$\Delta = -89.223$$

42.7%

$$J = \frac{7}{2}^+$$



$^{123}_{52}\text{Te}$

$$\Delta = -89.172$$

$$1.3 \times 10^{13} \text{ yrs (EC)}$$

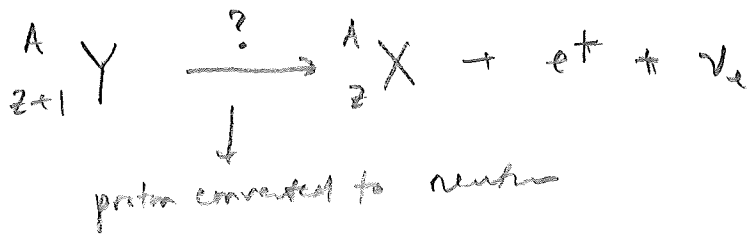
0.908%

$$J = \frac{1}{2}^+$$

β^+ decay

$\beta-16$

$$p \rightarrow n + e^+ + \nu_e \quad \text{not kinematically possible}$$



$$\begin{aligned} Q &= m_{\text{nuc}}(Y) - m_{\text{nuc}}(X) - m_e \\ &= [A u + \Delta(Y) - (z+1)m_e] - [A u + \Delta(X) - z m_e] - m_e \\ &= \Delta(Y) - \Delta(X) - 2m_e \end{aligned}$$

$$\text{If } \Delta(Y) - \Delta(X) > 2m_e = 1.022 \text{ MeV}$$

then Y is unstable to β^+ decay (also EC)

β^+ decay " followed by

$$e^+ + e^- \rightarrow \gamma + \gamma \quad (\text{pair annihilation})$$

$$Q = m_{e^+} + m_{e^-} = 1.022 \text{ MeV}$$

Each γ carries $\sim 0.511 \text{ MeV}$ (full-ble sign of e^+e^- annihilation)

[first observed 1934 Joliot-Curie; of Krane p.272]

Binding energy of a nucleus ${}^A_Z X$

$$\begin{aligned}
 B &= Z m_p + N m_n - m_{\text{nuc}}(X) \\
 &= Z [m({}^1\text{H}) - m_e] + N m_n - [m(X) - Z m_e] \\
 &= Z [1u + \Delta({}^1\text{H})] + N [1u + \Delta(n)] - [A u + \Delta(X)] \\
 &= Z \underbrace{\Delta({}^1\text{H})}_{7.289} + N \underbrace{\Delta(n)}_{8.071} - \Delta(X)
 \end{aligned}$$

Rewrite this in terms of $N+Z$ and $N-Z$

$$B = (N+Z)(7.680 \text{ MeV}) + (N-Z)(0.391 \text{ MeV}) - \Delta(X)$$

For ${}^{12}\text{C}$, $N=Z$ and $\Delta=0$

[B and Δ not the same]

$$B({}^{12}\text{C}) = 12 (7.680 \text{ MeV})$$

If we assume ${}^{12}\text{C}$ is "typical" so $N \approx Z$ and $\Delta \approx 0$

$$B = (7.7 \text{ MeV}) A$$

so binding energy per nucleon is approximately constant

[show next page] ⇒ "curve of binding energy"
[need to explain deviations]

$$\begin{aligned}
 m_{\text{nuc}}(X) &= Z m_p + N m_n - B \\
 &\approx (939 \text{ MeV}) A - (8 \text{ MeV}) A \\
 &\approx (931 \text{ MeV}) A \\
 &\quad \uparrow \\
 &\quad \approx 1u, \text{ chosen so } \Delta \approx 0
 \end{aligned}$$

Nuclei are ~1% less massive than sum of constituents
[recall deuteron about 0.1% → relatively weak binding]

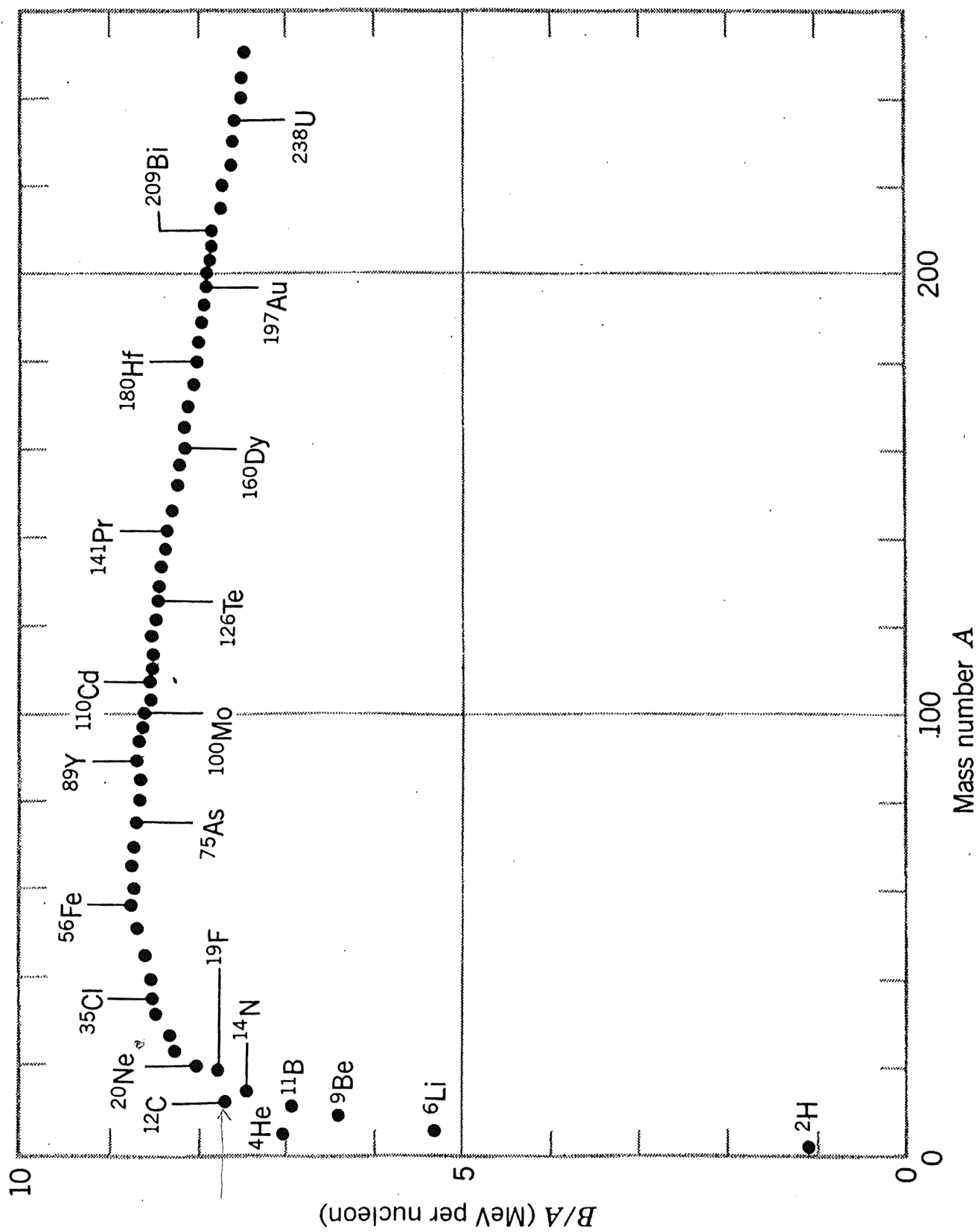


Figure 3.16 The binding energy per nucleon.

B/A for most abundant isotopes

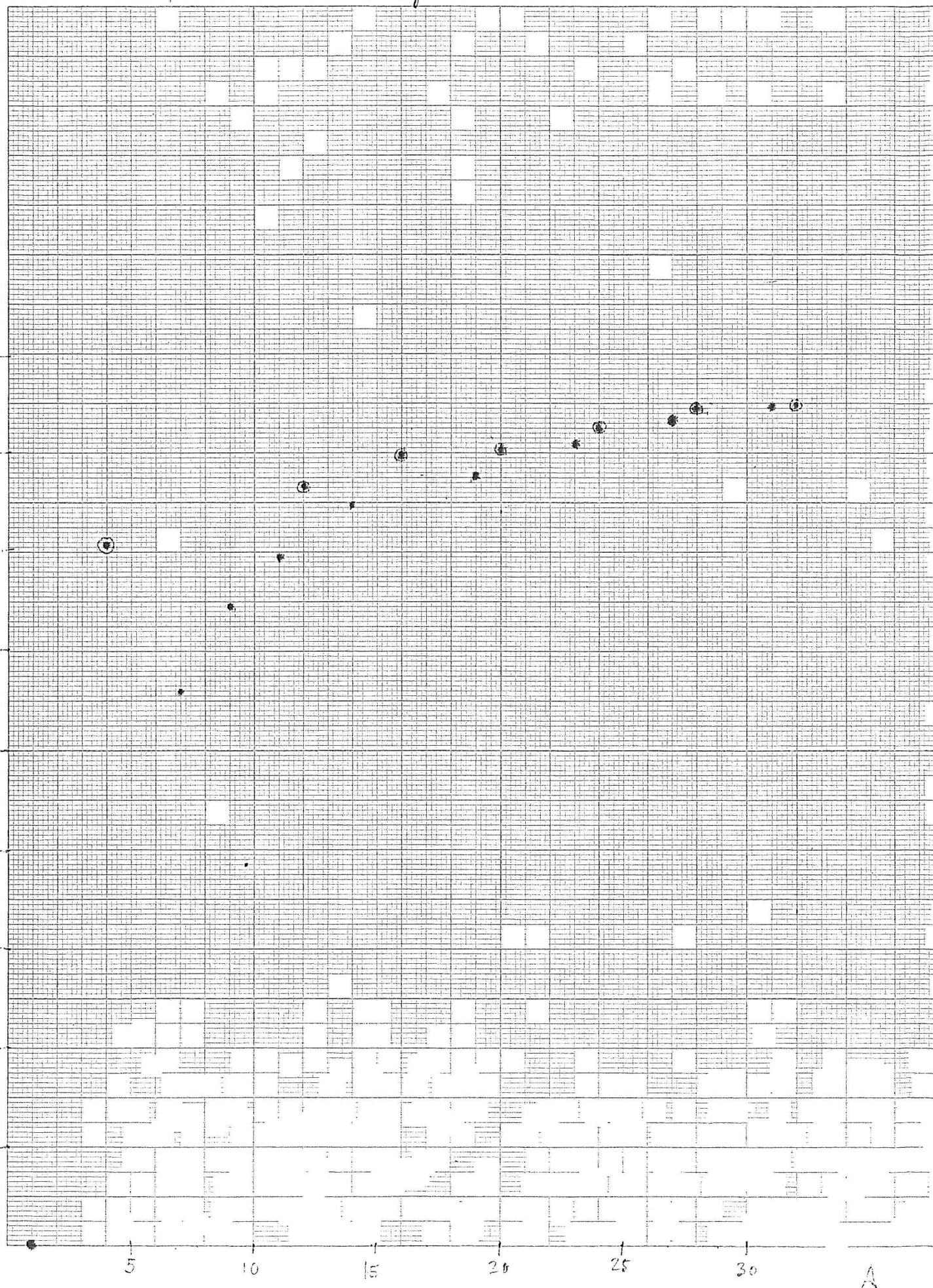
⊙ = even-even nuclides

EUGENE DIETZGEN CO.
MADE IN U. S. A.

$\frac{B}{A}$

NO. 340DR-MP DIETZGEN GRAPH PAPER
MILLIMETER

9
8
7
6
5
4
3
2
1



A

	Z	A	Δ	$A - 2Z$	$\frac{B}{A} = 7.68 + \frac{(A - 2Z)(0.39) - \Delta}{A}$
H	1	1	7.29	-1	0
He	2	4	2.42	0	[7.07]
Li	3	7	14.91	1	5.61
Be	4	9	11.35	1	6.46
B	5	11	8.67	1	6.93
C	6	12	0	0	[7.68]
N	7	14	2.86	0	7.48
O	8	16	-4.74	0	[7.98]
F	9	19	-1.49	1	7.78
Ne	10	20	-7.05	0	8.03
Na	11	23	-9.53	1	8.11
Mg	12	24	-13.93	0	[8.26]
Al	13	27	-17.20	1	8.33
Si	14	28	-21.49	0	8.45
P	15	31	-24.44	1	8.48
S	16	32	-26.81	0	8.49