

Nuclear transmutation

Rutherford & collaborators used α particles emitted by Po & Ra
 typical $T = 5 \text{ MeV}$ to probe atomic structure.

For heavy targets (Z_2 large), Coulomb barrier prevents
 α from getting too close to nucleus,
 which therefore behaves as a point charge

$$\frac{d\sigma}{d\Omega} \sim \frac{1}{\sin^4 \frac{\theta}{2}}$$

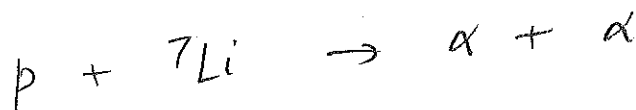
Scattering from lighter nuclei begins to depart from this
 suggesting a finite size for nucleus.

If α gets close enough ($\sim 1 \text{ fm}$), strong force
 can cause some of its particles to be absorbed

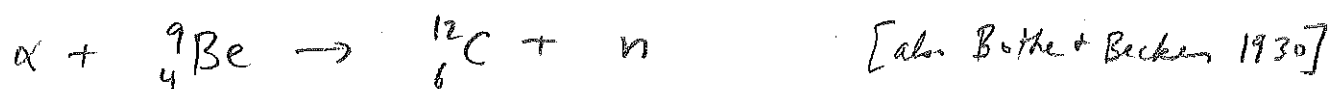
Nuclear transmutation (Rutherford, 1919)



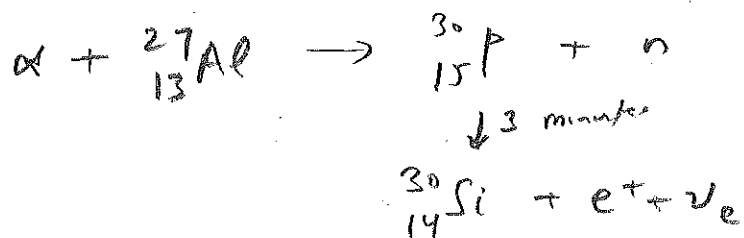
Cockcroft/Walton designed a linear accelerator to
 accelerate protons to $\sim 0.5 \text{ MeV}$.
 With this, they "split the atom" (1932)



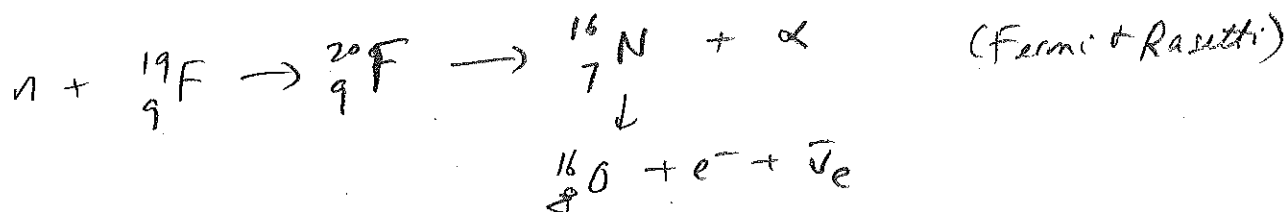
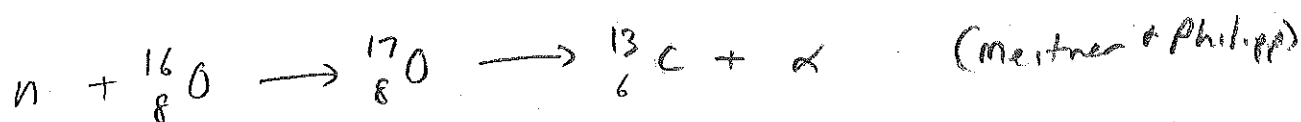
Chadwick discovers the neutron (1932)



Artificially radioactive isotopes (Joliot + I. Curie, 1934)

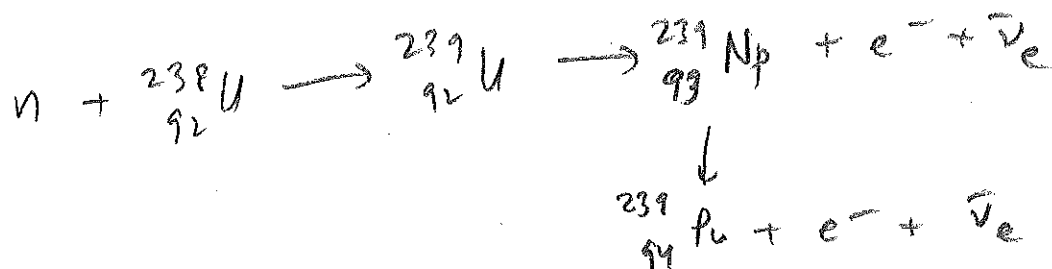


Neutrons as probes [no Coulomb barrier]



Many artificially radioactive isotopes produced
[nothing below fluorine]

even beyond uranium





It was found that for medium energy neutrons ($T \approx 20 \text{ MeV}$) the absorption cross section

$$\sigma \sim \pi R_x^2$$

Suggests most neutrons are absorbed by a nucleus they strike

Recall $R_x \sim A^{1/3} r_0$ where $r_0 = 1.2 \text{ fm}$

$$\sigma \sim A^{2/3} \pi r_0^2$$

$$\text{Uranium } (A=238) \Rightarrow \sigma \sim (13 \text{ fm})^2$$

Nuclear physicists define

$$1 \text{ barn} = (10 \text{ fm})^2 = 100 \text{ fm}^2 = 10^{-28} \text{ m}^2$$

"breadside of a barn"

uranium $\sim 1.7 \text{ barns}$

For higher energies, $\sigma \neq$ $\Rightarrow$ not all neutrons are absorbed
"partial transparency"

[cf graph from Evans]

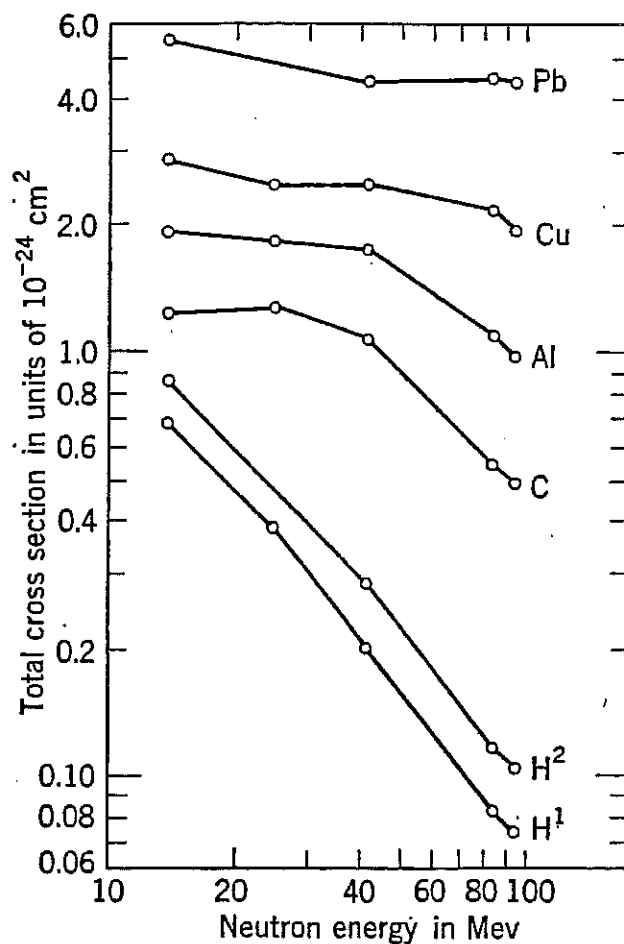
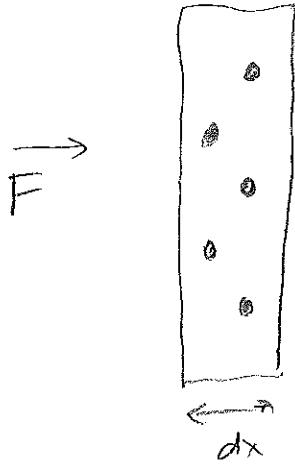


Fig. 2.3 Total neutron cross sections σ_t of five elements for neutrons of 14, 25, 42, 84, and 95 Mev. The decrease in σ_t at the higher neutron energies suggests that all neutrons which strike the target area $\pi(R + \lambda)^2$ are not captured and that nuclear matter is slightly transparent to swift neutrons. [From Hildebrand and Leith (H51).]

Let incident flux of neutrons be F



Probability of absorption by nuclei in film of thickness dx

$$f = n \sigma(dx)$$

$n = \# \text{ density of nuclei}$

$\Rightarrow$ Flux correspondingly diminishes

$$dF = -(n \sigma dx) F$$

$$\frac{dF}{F} = -n \sigma dx$$

$$\ln F = -n \sigma x + \ln F_0$$

$$F = F_0 e^{-n \sigma x} = F_0 e^{-\frac{x}{l}}$$

$l = \frac{1}{n \sigma} = \text{mean free path of neutrons}$

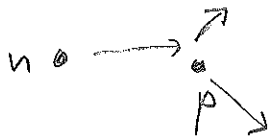
$$n = \frac{\rho}{A} N_A \quad \rho = \text{mass density}$$

$$U \rightarrow \rho = 19 \frac{\text{g}}{\text{cm}^3} \Rightarrow n = 4.8 \times 10^{28} \text{ m}^{-3}$$

$$\sigma = 1.7 \times 10^{-28} \text{ m}^2$$

$$l \approx 0.12 \text{ m}$$

Neutrons, being neutral, cannot be accelerated or stored
but they can be slowed down by collisions, or by
absorption + re-emission



Fermi found that absorption $\sigma \uparrow$ for lower energy n

[see graph for nickel]

Cadmium ~ 2000 barns for slow neutrons

[empirically $\sigma \sim \frac{1}{v}$]

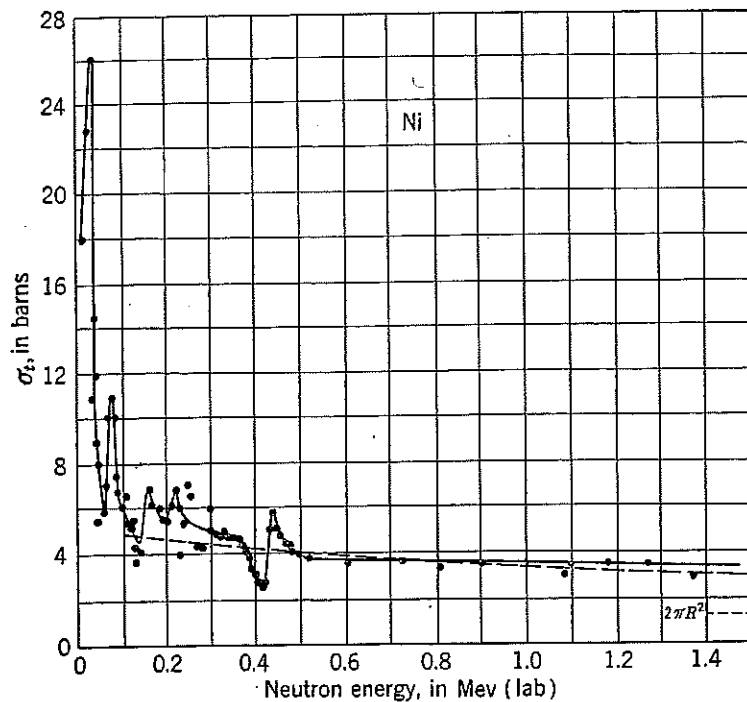


Fig. 2.1 The total cross section (absorption plus scattering) of nickel for neutrons from 0.01 to 1.5 Mev. Approximately monoenergetic neutrons (± 20 kev below 0.5 Mev; ± 150 kev above 0.5 Mev) were produced by bombarding a thin Li target with protons accelerated by an electrostatic generator. The dotted line is from the continuum theory of Feshbach and Weisskopf (F49), assuming a nuclear radius of 4.6×10^{-13} cm. [The experimental data are from Barschall, Bockelman, and Seagondollar (B15).]

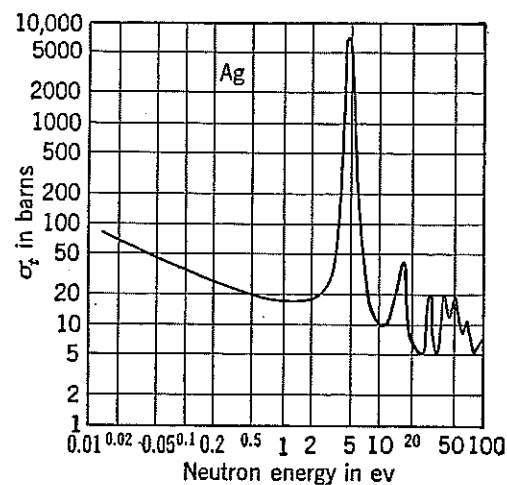
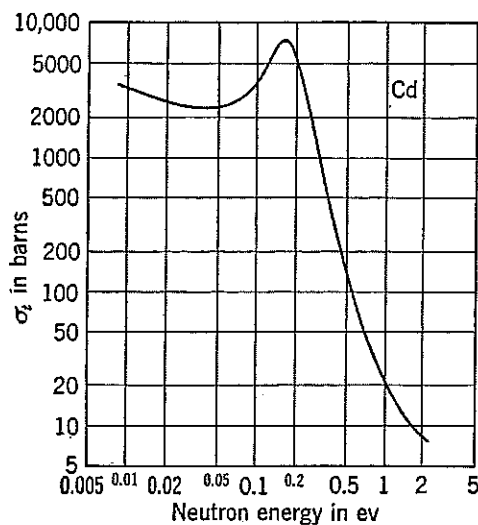
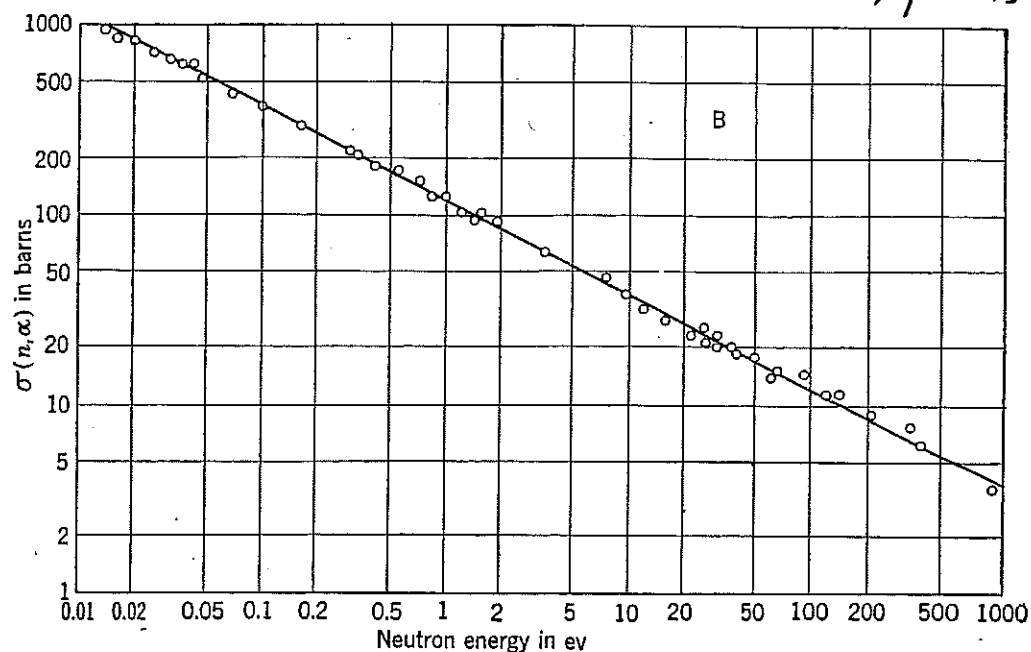


Fig. Introduction.1 Total neutron cross sections for B, Cd, and Ag, illustrating typical combinations of $1/V$ dependence (pure, in the case of boron) and slow-neutron resonances. The resonance in Cd at $E_0 = 0.176$ eV has a total width $\Gamma = 0.115$ eV and a maximum value of $\sigma_0 = 7,200$ barns. The curves are for the normal isotopic mixtures. In Cd, the 0.176-eV resonance is due entirely to Cd^{113} (relative abundance, 12.3 per cent); hence $\sigma_0 \approx 57,000$ barns for Cd^{113} alone. The complex γ -ray spectrum leading to the ground level of Cd^{114} has been measured by Bartholomew and Kinsey (B16). [From Goldsmith, Ibsen, and Feld (G29) and Hughes (H68).]

[skipped in 2019]

fr-6

why? wave properties of neutrons

de Broglie wavelength $\lambda = \frac{h}{p}$

Define reduced wavelength $\bar{\lambda} = \frac{\lambda}{2\pi} = \frac{h}{p}$

Consider non-relativistic neutrons ($T \ll m_n c^2 \sim 1000 \text{ MeV}$)

$$T = \frac{p^2}{2m} = \frac{1}{2m} \left(\frac{h}{\bar{\lambda}} \right)^2 = \frac{1}{2mc^2} \left(\frac{hc}{\bar{\lambda}} \right)^2 = \frac{1}{2(1000 \text{ MeV})} \left(\frac{200 \text{ MeV} \cdot \text{fm}}{\bar{\lambda}} \right)^2$$
$$= (20 \text{ MeV}) \left(\frac{1 \text{ fm}}{\bar{\lambda}} \right)^2$$

$$\frac{\bar{\lambda}}{1 \text{ fm}} = \sqrt{\frac{20 \text{ MeV}}{T}}$$

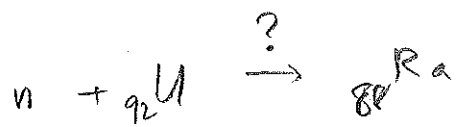
For $T \gtrsim 20 \text{ MeV}$, $\bar{\lambda} < 1 \text{ fm}$, neutrons act as a particle
($\bar{\lambda} \ll R$)

Small T , $\bar{\lambda} \gg R$, act like a wave, $\sigma \propto \bar{\lambda}^2$

Neutrons of $T \sim \frac{3}{2} kT \sim 0.05 \text{ eV}$ are called "thermal neutrons"

1938 Hahn & Strassmann (chemists)

thought they observed



but no physical process could explain this

- why did they think it was radium?
- Because it had similar chemical properties to Ba and a radioactive product was precipitating out of soln w/ Ra
- But they then tried to separate Ra from Ba and failed
- What did they do? Consulted a physicist.

Former colleague Lise Meitner

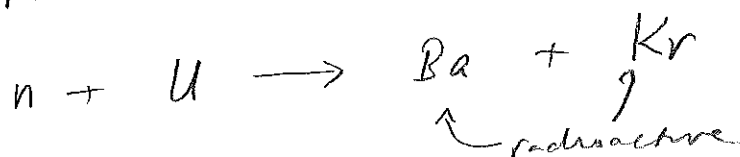
an Austrian Jew.

When Hitler annexed Austria, she lost her citizenship

In 1938, she escaped to Sweden.

She discussed it w/ nephew Otto Frisch and decided it was Ba.

1939 Meitner & Frisch propose



"nuclear fission"

Why didn't anyone think of this before?

$$\frac{B}{A} \text{ for U} \sim 7.5 \text{ MeV}$$

$$\frac{B}{A} \text{ for Ba, Kr} \sim 8.2 \text{ MeV}$$

about 0.7 MeV released per nucleon

$\Rightarrow \sim 170 \text{ MeV per fission.}$

$\Rightarrow$ kinetic energy of decay products

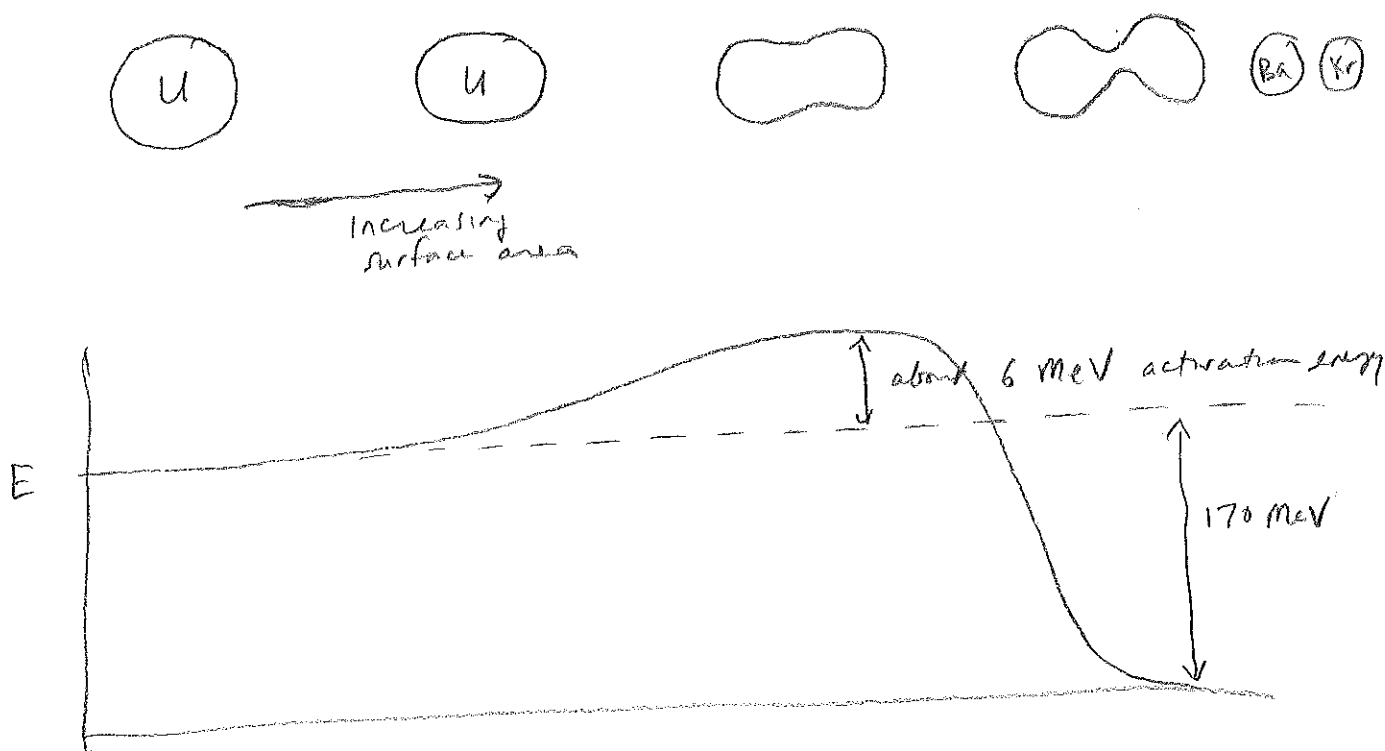
Frick in his lab verified energy of fragments.

fi-9

If U less stable than Ba, Kr why doesn't it just decay?

Think α -decay. Coulomb barrier holds them in.

Alternative picture. Liquid drop model (surface tension)



Tunnelling through barrier can occur but is rare
"spontaneous fission"

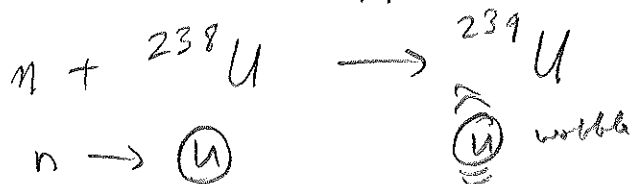
$$T_{\frac{1}{2}} \sim 3.5 \times 10^{17} \text{ years}$$

[HKR, p. 1169]

(α -decay is more likely)

Incident neutron supplies energy

fig-10



$$Q = \Delta({}^{238}\text{U}) + \Delta(n) - \Delta({}^{239}\text{U}) = 4.8 \text{ MeV}$$

$$\begin{bmatrix} 47.306 \\ 8.071 \\ -51.571 \\ \hline 4.806 \end{bmatrix}$$

energy released causes drop to wobble
but activation energy of ${}^{239}\text{U}$ is 6.6 MeV

[? 5.7 MeV in
HRK, 1156]

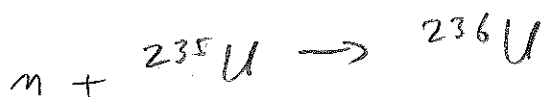
neutrons must have $E > 1.8 \text{ MeV}$
to initiate fission

"fast neutron fission of ${}^{238}\text{U}$ ", $\frac{v}{c} \sim \sqrt{\frac{2T}{mc^2}} \sim 0.06$

But Hahn & Strassmann were using slow (thermal) neutrons
($T < 1 \text{ eV}$)

1939 Bohr realized that they were observing
"slow neutron fission of ${}^{235}\text{U}$ "

${}^{238}\text{U}$ 99.3%
 ${}^{235}\text{U}$ 0.7%



$$Q = 6.5 \text{ MeV}$$

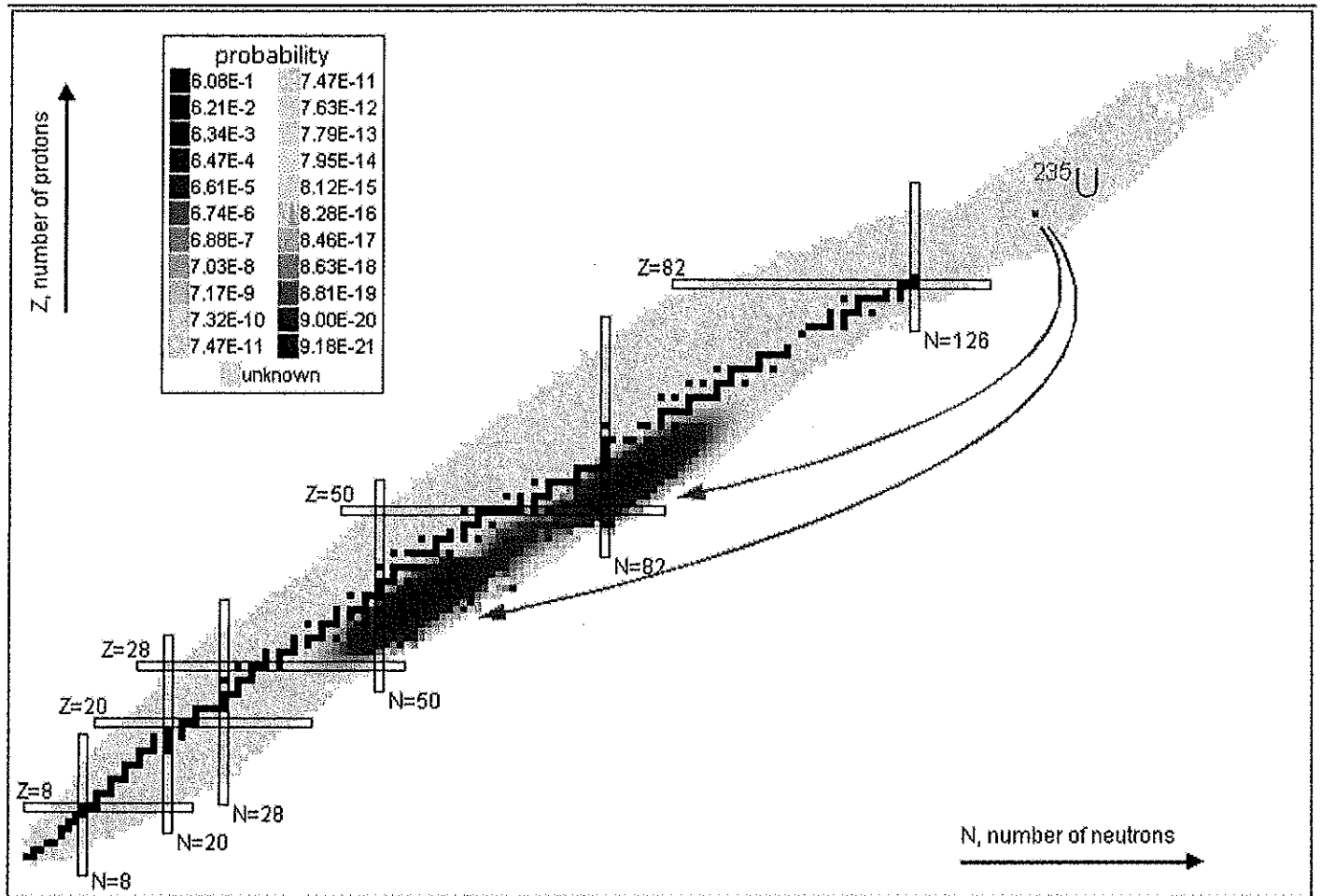
⇒ why difference? even → odd
odd → even

activation energy of ${}^{236}\text{U}$ is 6.2 MeV

no additional kinetic energy required!

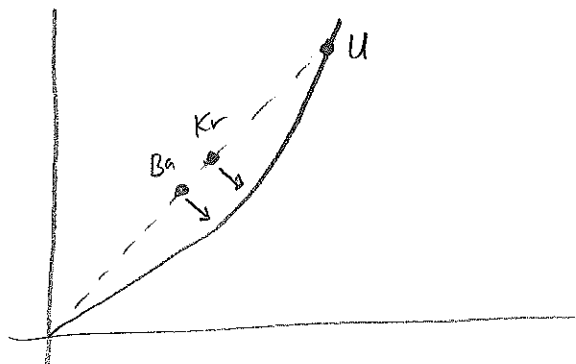
$$\begin{bmatrix} 40.916 \\ 8.071 \\ -42.442 \\ \hline 6.545 \end{bmatrix}$$

[? 5.2 in HRK]



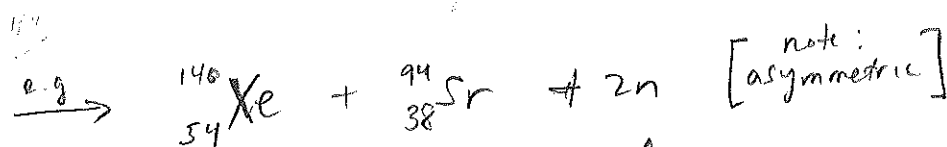
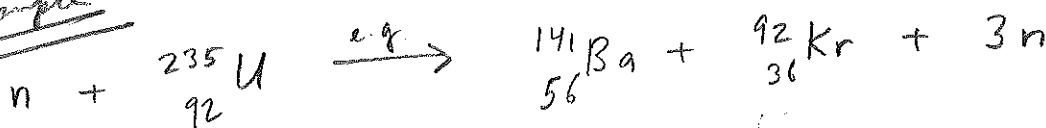
Lab Deming Lounge ~~Hyper~~
blogspot.com

[Why are Ba + Kr radioactive? Because they are neutron rich]



Fission often releases 2 or 3 neutrons in addition to fission fragments

For example



Released neutrons can initiate further fissions

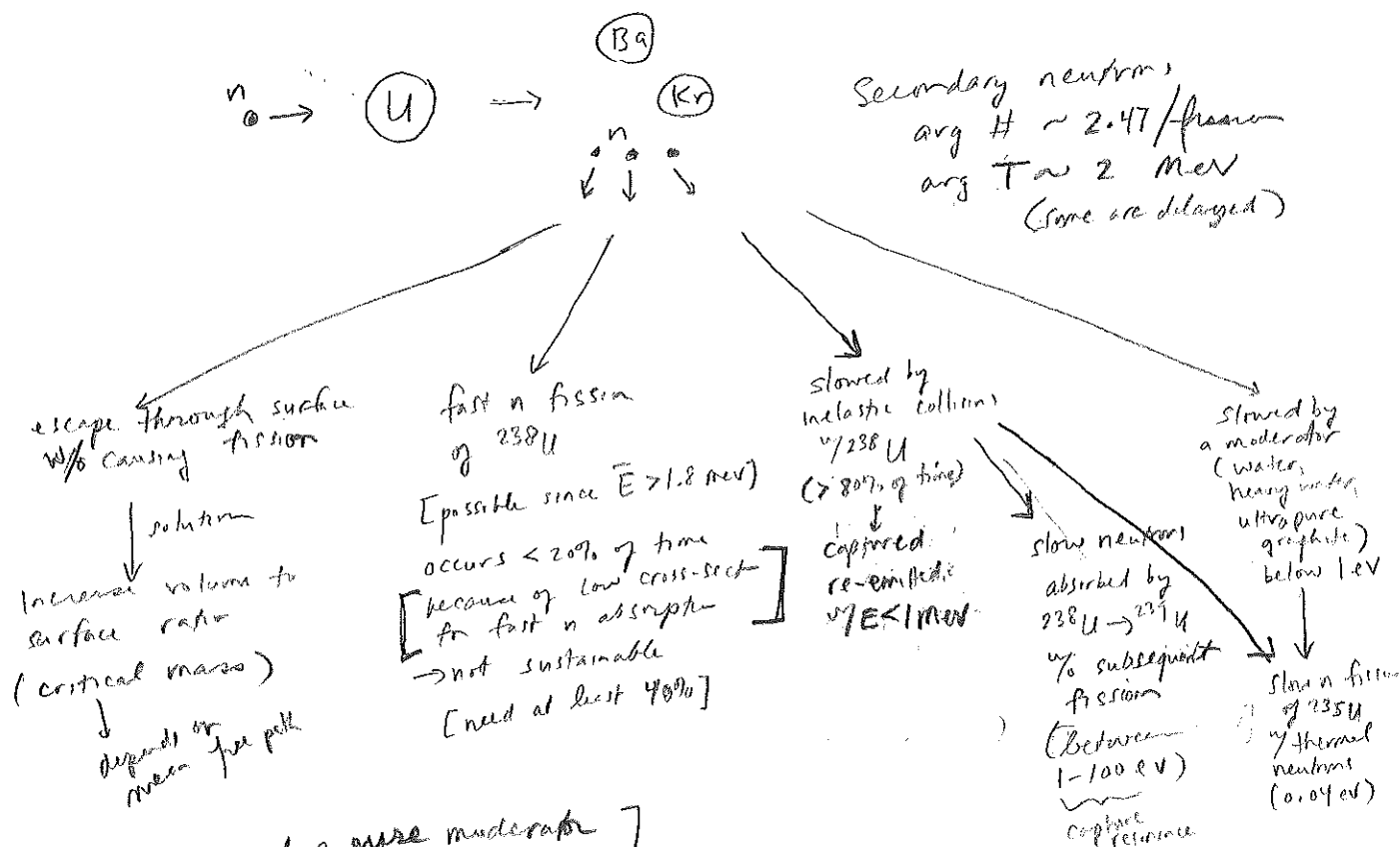
$\Rightarrow$ chain reaction

controlled $\Rightarrow$ nuclear reactor
uncontrolled $\Rightarrow$ weapons

about 110 MeV released per fission

vs 1-10 eV released in chemical reactions $\Rightarrow$ factor of 10^7 to 10^8

First controlled nuclear reaction
12/2/1942 U of Chicago squash court (Fermi)



[Need: critical mass, and a pure moderator]

first k = multiplication factor = # secondary neutrons that go on to produce subsequent fissions

$k < 1$ subcritical

$k = 1$ critical (self-sustaining)

$k > 1$ supercritical

$$P(t) = P_0 k^{\pm \frac{t}{\tau_{\text{gen}}}}$$

use neutron absorbers to keep reaction from going supercritical
eg Cadmium control rods

[Richard Rhodes, Making of the Atomic Bomb]

Just talk about this:

fr-13

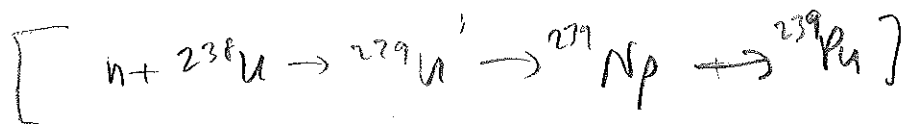
[What it would take a supercritical rxn, ie a bomb?]

- slow n fission not suitable, energy release causes U to expand & $\therefore$ n's will all escape
- fast n fission of ^{238}U impossible because $k < 1$.
- fast n fission of ^{235}U is possible
- enrichment:

electromagnetic separation	} Oak Ridge
gaseous diffusion	

Little Boy ~ 80% ^{235}U (about 50 kg)

(critical mass ~ 10cm)



^{239}Pu is fissionable

[Fat Man]

Nuclear physics

Mentzer biography (Sime)
(also Philip Ball: saving the bank)

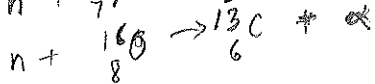
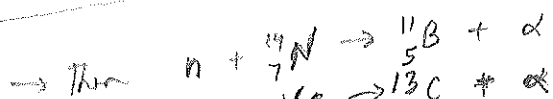
1949 Rutherford induced nuclear rxns w/ α



but not until 1924 was it reborn. It was abashed, not scattered.
(Blackett). The 1920s $\Rightarrow$ existence of a strong attractive force that
overcame Coulomb repulsion.

Nuclei heavier than K never absorbed α 's, however, nor did Li, Be, C, O.

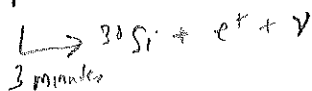
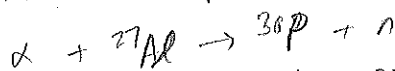
→ Bief (1930) Bothe & Becker discovered ^{a new penetrating type γ} radiation from α (from Po) bombardment of Be, Li, B. . Joliot & Irene Curie (1932) found it to radiate ejected p from paraffin, & interpreted it as p, but Chadwick proved it was really neutrons (1932)

$$\alpha + {}^9\text{Be} \rightarrow {}^{12}\text{C} + n$$


was derived by Norman Flecken
" " " Matheson & Phillips

1934 Fermi theory of β -decay, which used Pauli's ν idea

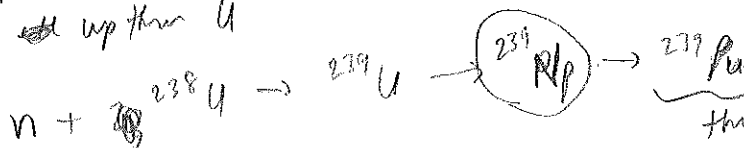
1934 Tolson & Curre produced artificial radio-chlorz



19391 Female ~~large~~ + Paleth Bombarded etc 7 n
+ dark radiating 2 F. 1

producing ortho and para isomers of F, Al, Si, P, Cu, Zn.

all up there U

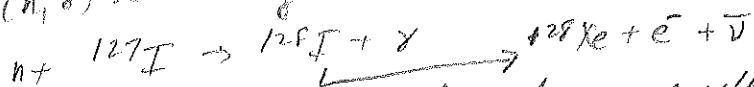


this was hypothesized
but not proved

Fermi observed (n, α) rxns & (n, p) rxns on lighter elts



and (n, γ) rxns on heavier



and all the radioactive nuclei always emitted e^-

~~Maria + W. decided to start studying transmutations~~

Szold + Chalmers observed



~~that~~ (lower energy)
Maria found that n could not be absorbed by Al, Na, Si
which required higher energies to split off an α or p
but only in capture process by heavier nuclei Ag, I, Au

She suggested that slower neutrons are more easily absorbed
than fast neutrons

1934 Fermi showed that paraffin, etc. slowed the n , after which they
were more easily absorbed by Ag, Cu, I, ~~etc.~~

Uranium enrichment

meth.
 process {
 → electromagnetic separation
 → gaseous diffusion

→ large ~~gas~~
 centrifuges (most common
 + economical)

3-5% ^{235}U → power reactors

70-90% ^{235}U → weapons

Hiroshima → 64 kg of ^{235}U ⇒ 15 kilotons
 could do it w/ 25 kg (size of a melon)

Trinity → 6 kg of ^{239}Pu ⇒ 20 kilotons
 could do it w/ 5 kg or less (size of a plum)